

Department of Natural Resources  
MARYLAND GEOLOGICAL SURVEY  
Jeffery P. Halka, Acting Director

**COASTAL AND ESTUARINE GEOLOGY  
FILE REPORT NO. 10-02**

**SEDIMENTATION ANALYSIS OF NEW GERMANY  
LAKE**

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Prepared For

**Department of Natural Resources**

**APRIL 2009**

Revised November 2009

## **EXECUTIVE SUMMARY**

In response to a request by the Department of Natural Resources, Maryland Park Service and Engineering and Construction, Maryland Geological Survey (MGS) was charged to study the sedimentation of New Germany Lake located in Garrett County in the State of Maryland.

Sediment cores were collected from the lake, sediment accumulation volumes were determined and physical and chemical characteristics of the sediment were analyzed.

Sediment cores were collected in October 2008. The cores were analyzed and a sediment accumulation thickness ranging from 0.56 meters to 1.16 meters [1.8 to 3.8 feet] was observed throughout the cores. The calculated amount of sediment accumulated since the construction of New Germany Lake is currently a maximum of 33,191 cubic meters [43,412 cubic yards] within the confines of the current shoreline.

Four cores were analyzed for historical patterns in lead and zinc. Three of these cores displayed patterns that suggest only a portion of the sediment has been deposited since the construction of the New Germany Lake Dam and the remaining sediment is from the original Swauger's Mill dam.

An elemental analysis was performed on thirty sub-samples from the sediments in the collected cores. The only elements which showed enrichment above national screening levels were aluminum, iron, and barium. However, the elevated levels of these elements are most likely due to the analysis methods which were used.

Three samples were analyzed for priority pollutants using EPA 8270 methods. These samples all reported results below the detection limits for the laboratory equipment.

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New Germany State Park  
Historic Shoreline and Core Locations  
October 2008

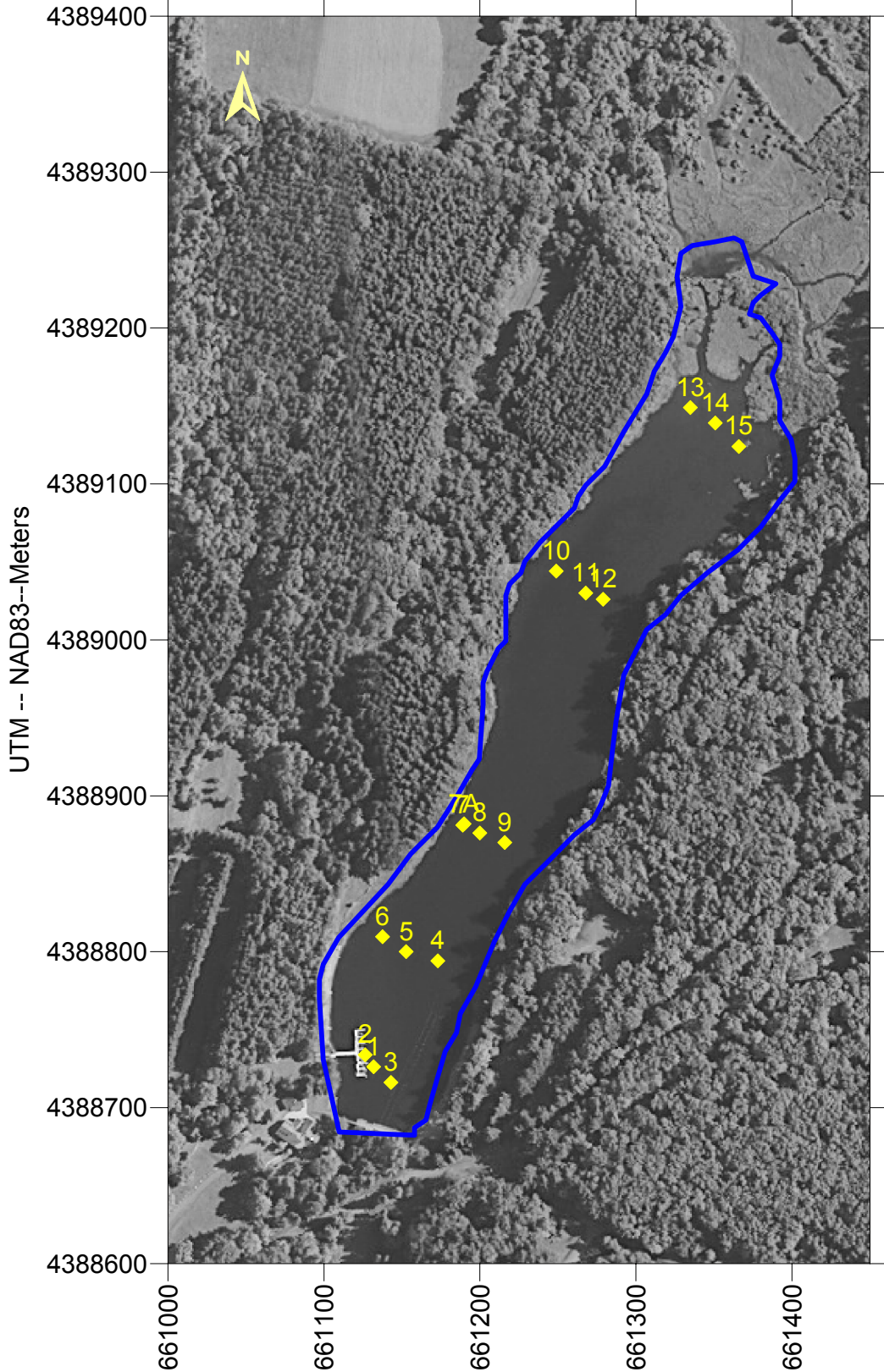


Figure 1. New Germany Lake. Aerial Photography is from 2007 NAIP imagery. Blue Shoreline is digitized from USGS 1946 Topography Maps. Yellow Symbols indicate sediment cores collected in this study.

## **SITE DESCRIPTION**

New Germany Lake is a manmade twelve acre impoundment on Poplar Lick Run in New Germany, Maryland. Upstream and downstream reaches of the contributing and surrounding waterways suggest that this site was once a channelized stream approximately 3.0 to 4.6 meters [10-15 feet] in width and 0.3 to 0.6 meters [1-2 feet] in depth. Around 1837-1847, a milldam and resulting millpond were constructed at what appears to be the same site as the current dam (USGS 1899, 1904, 1938 topographic maps). That dam was lower in height than the current dam; however, it still created a pond of approximately 9 acres (USGS 1899, 1904 topographic maps). Poplar Lick Run enters the lake at the north, and continues southward to the Savage River upon exiting the lake at the dam located on the south end of the lake. The marsh located at the north end of the lake is likely caused by the impoundment and was not originally found in that area (MGS, 1902). Without bathymetry and accurate topography from that time, it is impossible to determine the original depth of this initial impoundment. In the 1933-1935 timeframe, a new earthen dam was constructed which raised the level of the lake to its current height. The lake was very rarely lowered after its final construction in the 1930s. In 2008, DNR performed some safety upgrades and maintenance on the dam which did not alter the water level of the lake. While performing these upgrades, it was noted that the drain pipe for the dam was still clearly above the sediment level. While no evidence has been collected that documents the growth of the headwaters marsh, a cursory observation shows a dendritic stream network with spatially equal amounts of water as there was "land". The streams in this marsh appear to be fed as much by springs as from surface water, and the channels are 0.75 to 0.90 meters [2.5-3 feet] deep with coarse grained sediments throughout. Except for the manmade beach area, the shorelines of the lake do not appear to be marked by much erosion.

There are four known anthropogenic activities on the lake which may have influence over the sediments found in the impoundment. Timbering operations were prevalent throughout the area surrounding Poplar Lick Run prior to, during, and after the construction of the dam. Additionally, a majority of the land located immediately west of New Germany Lake was being used for agriculture. These operations would have increased the flow of sediment into the drainage basins. There also was a sawmill located approximately 400 meters [437 yards] upstream from the current lake's headwaters. This sawmill was in operation beginning sometime around 1800 and ceased operations in the 1950s. Evidence of sawdust was in the reconnaissance cores collected in September 2008. Finally, the lake has been used as a recreational facility since 1935. The largest sediment impact from this activity is the import of sand to the current and historical manmade beach areas.

## **GEOLOGIC BACKGROUND**

New Germany Lake occupies a basin developed in the Pocono Formation. The lake's western shore roughly traces the contact between the Pocono and the Greenbrier Formations, at the base of Meadow Mountain. Further upslope on Meadow Mountain, above the Greenbrier Formation, are exposures of the Mauch Chunk Formation, capped by the Allegheny Formation and Pottsville Group. The sedimentary sources to the New Germany Lake watershed are primarily confined to these four geologic units.

The lake is surrounded on the north, east and south sides by soils developed in the Pocono Formation, which consists of sandstone, siltstone, shale and some conglomerate. The Pocono typically weathers to a grayish, sandy soil to a sandy loam, containing fragments of

sandstone and conglomerate cobbles. Pocono soils usually range from 0.38 to 0.61 meters [15-24 inches] in depth and stone fragments and boulders are common locally. The Greenbrier and Mauch Chunk Formations are similar in composition, consisting of shale and sandstone, although the Greenbrier is more calcareous, containing a limestone member. Along hill slopes these shales and limestones weather to a heavy red loam or clay with abundant sandstone boulders. Heavy, yellow and brown clay soils develop in valleys underlain by the Greenbrier and Mauch Chunk Formations. Because these formations lay upslope from the lake valley, they would be expected to contribute red and yellow clays, sand, and rock fragments to the lake's sediment burden. Similar materials would be expected to be derived from the Allegheny Formation and Pottsville Group which make up the western hilltops along the lake. (MGS, 1902)

## **STUDY OBJECTIVES**

The objectives for this study were:

- 1) Determine the sediment accumulation within the New Germany State Park Lake.
- 2) Document the physical and elemental characteristics of the sediment.
- 3) Identify any priority pollutants which may be contained in the sediment.

## **METHODOLOGY**

### *Core Collection*

Cores were collected along five transects of the lake and one core on the downstream river bank. The lake cores were spaced so that one core corresponded to the center of each transect and the two other cores corresponding to the halfway points between the shorelines and the the centerpoint of each transect. Coring locations are shown and documented in Figure 1 and Appendix A. Sediment cores were collected in aluminum liners attached to a vibracore head supplemented with 113 kilograms [250 pounds] of added weight. Cores were driven to refusal, capped, labeled, and retrieved. Horizontal control was provided through a Thales Navigation ProMark 3 GPS supplemented with satellite based augmentation system (SBAS) differential corrections providing a real-time horizontal accuracy of 2-5 meters [6-15 feet]. Horizontal positions were recorded in the Universal Transverse Mercator (UTM) system based upon the North American Datum of 1983 (NAD83). Core logs include depth of water, total depth driven, and GPS coordinates of each core. All cores were collected between October 6, 2008 and October 10, 2008.

### *Sediment Documentation and Sampling*

Cores were drained, split, and documented within 24 hours of collection. Each core was drained of water to the sediment-water interface and then cut in half lengthwise to expose the collected sediments. The sediments were characterized and documented in sediment logs and photographs. These sediment logs are in Appendix B. The photographs are maintained separately by the Maryland Geological Survey. The cores were sub-sampled for further laboratory analysis with sampling intervals annotated on the core logs and on the sample collection bags. Samples collected for physical property analyses and elemental analyses were collected at discrete intervals within the cores where physical changes were identified. Samples collected for priority pollutants were collected equally along the length of the deposited sediment. All samples were kept in a cooler to maintain a temperature between 8-14° C [46-57° F] .

### Physical Property Analyses

Selected sub-samples were analyzed for water content, bulk density, and grain size (gravel, sand, silt, clay contents). Analyses were performed as soon as possible after sample collection, and all samples were refrigerated in sealed Whirl-Pak™ plastic bags prior to analysis.

In preparation for grain size analysis, sediment samples underwent a cleaning process to remove soluble salts, carbonates, and organic matter. These constituents may interfere with the dispersal of individual sediment particles and, thereby, affect the subsequent separation of the sand and mud fractions. All sediment samples were treated first with a 10% solution of hydrochloric acid (HCl) to remove carbonate material, such as shells, and then with a 6% or 15% solution of hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>) to remove organic material. A 0.26% solution of the dispersant sodium hexametaphosphate ((NaPO<sub>3</sub>)<sub>6</sub>) was then added to ensure that individual grains did not clump, or flocculate, during pipette analysis.

For each sample, the coarse fraction was separated from the mud fraction by wet-sieving through a 4-phi mesh sieve (0.0625 mm, U.S. Standard Sieve #230). The sand fraction (i.e., particles > 0.0625 mm) was dried and weighed. The mud fraction (i.e., sediment passing through the #230 sieve) was analyzed using a pipette technique to determine the proportions of silt and clay (Krumbein and Pettijohn, 1938). The mud fraction was suspended in a 1000-ml cylinder in a solution of 0.26% sodium hexametaphosphate. The suspension was agitated and, at specified times thereafter, 20 ml pipette withdrawals were made. The rationale behind this process is that larger particles settle faster than smaller ones. By calculating the settling velocities of different sized particles, withdrawal times can be determined. At the time of withdrawal, all particles larger than a specified size have settled past the point of withdrawal. Sampling times were calculated to permit the determination of the total amount of silt and clay (4 phi) and clay-sized (8 phi) particles in the suspension. Withdrawn samples were dried at 60°C and weighed. From the dry weights, the percentages of gravel, sand, silt, and clay were calculated for each sample and classified according to Shepard's (1954) nomenclature, or, if sample contained gravel, according to Folk's (1954) nomenclature. Shepard's classification, which is widely used in sediment studies, is based on the relative percentages of the sand, silt and clay components of the sediment. Sediments are classified as one of 10 classes according to Shepard's ternary diagram (Figure 2). Folk's classification of sediments is based on textural composition of gravel, sand and mud (silt + clay) fractions and consists of 15 classes. (Figure 3).

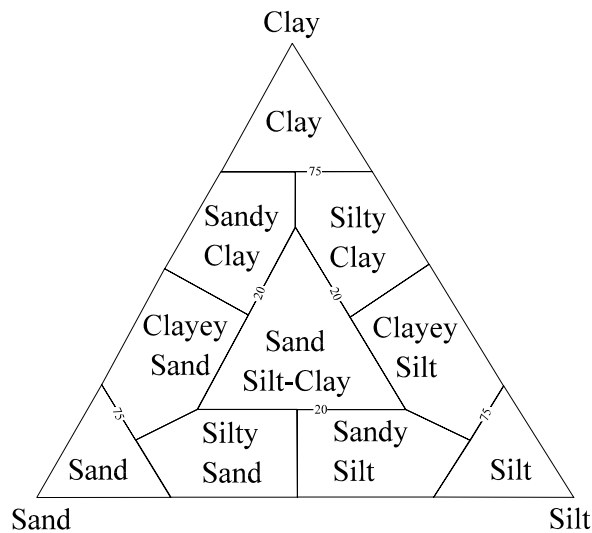


Figure 2. Shepard's (1954) classification of sediment types.

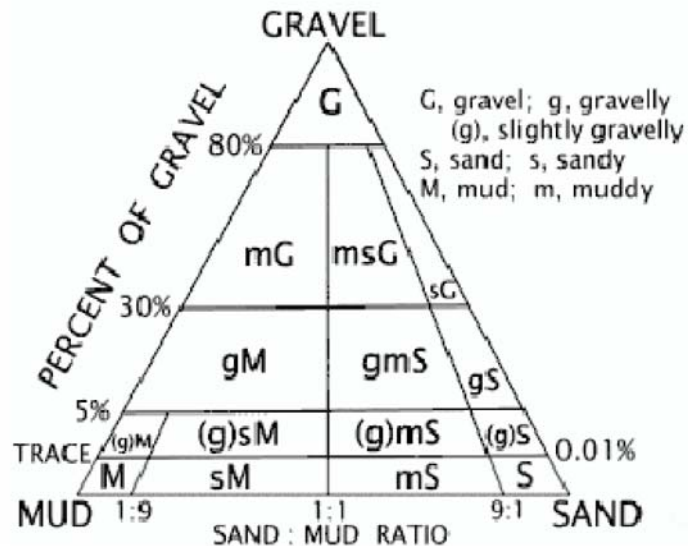


Figure 3. Folk's (1954) classification of sediments.

Elemental Analyses

Activation Laboratories, Ltd. (Actlabs) of Ancaster, Ontario, Canada, analyzed the sediment samples for 48 elements. Concentrations of the elements were determined by one or both of two methods: Instrumental Neutron Activation Analysis (INAA) and a Total Digestion – ICP Analysis. For the Total Digestion method, Actlabs used a four-acid, “near total” digestion process, followed by analysis of the digestate by inductively coupled plasma emission spectroscopy (ICP-OES). The four-acid digestion employed perchloric (HClO<sub>4</sub>), hydrochloric (HCl), nitric (HNO<sub>3</sub>), and hydrofluoric (HF) acids. The digestion method is similar to EPA Method 3052 used for total decomposition of samples.



Quality assurance was checked using the method of bracketing standards (Van Loon, 1980). The standard reference materials (SRMs), similar to the sediments being analyzed, were included every 8 samples and submitted to Actlabs as blind unknowns. Actlabs' results of the analyses of the SRMs are listed in Appendix F.

The authors point out that the analytical method used in this study for the elemental determinations is not appropriate for regulatory applications that require the use of leachate preparations (*i.e.*, EPA Methods 3050, 3051, 1311, 1312, 1310, 1320, 1330, 3031, or 3040). This method is appropriate for those applications requiring a total decomposition for research purposes (*i.e.*, geological studies, mass balances, analysis of Standard Reference Materials) or in response to a regulation that requires total sample decomposition.

#### Priority Pollutant Analyses

Three collected core samples were homogenized and they were analyzed using EPA 8270 standards for priority pollutants. These samples were collected, stored in acid-washed glass containers with Teflon lid liners, and analyzed by the Maryland Department of Health and Mental Hygiene Laboratories in Baltimore, Maryland.

## **RESULTS AND DISCUSSION**

#### Sediment Accumulation

Sixteen cores were collected from New Germany Lake and one core was collected from the downstream river bank. The core logs, photographs, physical properties, and elemental properties were used to determine the core depth of various layers including historical soil, coarse grained deposits, accumulated fine grained sediments, and a dominant organic layer.

Identification of these selected layers was chosen to aid in the determination of accumulated sediment within the lake. The historical soil layer is defined through the dominant soil types and geology found in the local vicinity and confirmed through the collection of the downstream riverbank core. This layer is visibly and textually identifiable through its dark red sandy soil structure with included gravel, cobbles and rock conglomerates. There are no embedded organic materials and the water content is below twenty percent. Eight of the cores penetrated to the depth of the historic soil level. The coarse grained deposit layer is clearly identifiable through its physical characteristics. This layer is predominately yellow or grey in color consisting of a high percentage of sand and occasionally including organic material. This layer is believed to be a subaerial deposit formed by the weathering of the uphill sediments and their subsequent deposition on the valley floor through wind and rain events prior to any water inundation. This layer was observed in fifteen of the cores. The accumulated fine grained sediment layers, identified as soft mud and firm mud, are the materials which have collected since an impoundment of waters created a subaqueous sedimentary environment. The impoundment slowed the flow of waters and decreased the energy in the environment allowing the fine grain particles to fall out of the water column and collect on the bottom of the lake. These layers are clearly identified through the sediment's physical properties demonstrated by an increase in clay and silt percentages and an increase in water content due to its subaqueous formation. Furthermore, this layer is identified through the occasionally observed striations from episodic depositional events. Organic material is found throughout this layer. These sedimentary deposits were found in all of the cores within the lake. The last layer which was identified is a layer which was extremely high in organic material. In several cores, this layer included sawdust that was most likely deposited by a storm event relocating material from an

upstream sawmill. In other cores this layer was a dense leaf mat or a heavy accumulation of sticks. Identification of this layer was found in fifteen of the sixteen cores from the lake.

A description of the selected facies in the collected cores is best visually summarized in the graphs in Appendix C. Figure C-1 is a bar graph which depicts the location of these layers from the sediment surface throughout the depth of the collected cores. Figure C-2 is a bar graph which depicts the location of these layers from the surface of the lake which aligns all of the layers to the same vertical plane. Table C-1 presents the interpreted data from the cores.

A majority of the cores collected follow clear definable patterns of sedimentation. From the lake bottom downwards, the topmost sediments are mostly silts and clays with organic material deposited throughout. The sediments generally become more consolidated with depth. These silts and clays are the sediments which have collected since the initial impoundment of the waters. Below the silts and clays is a thin layer of coarse sand, ranging from 2-33cm [1-12 inches] in thickness. This layer of coarse sand is likely the deposits created by wind and rain erosion of the upslope soils which were mobilized and deposited into the valley prior to the creation of any pond or lake. Below the layer of coarse sand is the historical soil horizon.

Cores 8, 12 and 13 demonstrate a clear sedimentary pattern in recent sediments; however, at depth the sediments appear to have been reworked. From the lake bottom downwards, the topmost sediments are mostly silts and clays with organic material deposited in layers. These sediments become more consolidated with depth. Below the silts and clays there is a deposit of coarse grained material which varies greatly in thickness from at least 8-55 cm [3-22 inches]. The coarse grained materials are characterized by predominantly medium to coarse sands with pockets of organic material and mud lenses. The organic material and mud lenses are most likely due to these cores being near the headwaters of the initial millpond or even in the floodplain or meanders located upstream of the millpond or in the case of Core 8, it may have been in the historic stream channel. In all cases this deposit would be subject to reworking from storm events. The lack of fine sediments suggests that these coarse grained deposits were mobilized under high energy events which indicate that these are alluvium deposits.

A very dense organic layer was detected in all cores with the exception of core 5. This organic layer is likely a distinct event or series of events which occurred within the drainage basin. There is not enough data to draw any conclusions from this layer.

The accumulated sediment is spatially variable, dominantly dependent on the pre-existing topography of the valley and the current bathymetry of the lake. In general, there is an 81 cm [range: 67-93 cm] [31 inches] sediment deposition at the southerly end of the lake slightly decreasing to a 68 cm [range: 56-92 cm] [26 inches] sediment deposition in the middle of the lake and increasing to a 88 cm [range: 56-106 cm][34 inches] sediment deposition at the north end of the lake. The increase in sediment deposition at the north end of the lake also is identified with an increase in silt content of the deposited sediments. This is anticipated as silts need more energy to stay suspended in the water column than clay particles and therefore they deposit to a greater extent closer to the headwaters. This increase is also an indication of the natural pro-delta formation extending the headwaters marsh and decreasing the length of the lake.

Aerial imagery from 2007 maps the spatial area of New Germany Lake to be 41,700 square meters (10.3 acres). Using this area, the lake was divided into thirds, measuring upstream from the dam, and areas and volumes were calculated. While it is likely that the accumulated sediment varies spatially and tapers to a much thinner deposit near the shorelines, this calculation does not extrapolate those changes as the extent of deposit and rate of change is unknown. The calculations use the average accumulated sediment thickness within the area multiplying it by the planar area of the current lake. This will yield an overestimate of the actual accumulated sediment volume in each area; however, it will underestimate the total volume of sediment

collected as it does not factor in the sediment which has collected in the headwaters of the lake. Table 1 below presents this data. The total accumulated sediment in New Germany Lake within the current shoreline is 33,191 cubic meters [43,412 cubic yards].

Section	Distance from Dam (Meters)	Average Accumulated Sediment (Meters)	Area (Square Meters)	Calculated Accumulated Sediment Volume (Cubic Meters)
South	0-125	0.81 [0.67-0.93]	9500	7695
Middle	125-306	0.68 [0.56-0.92]	14200	9656
North	306-530	0.88 [0.56-1.16]	18000	15840
Total	0-530	-----	41700	33191

**Table 1. Sediment accumulation within New Germany Lake.**

### Elemental Analyses

Analytical results are presented in Table F-2.

Because the samples were analyzed using a total decomposition method (four-acid digestion), the concentration values should not be compared to threshold limits in the NOAA SQuirTs (Screening Quick Reference Tables) (Buchman, 2008). The values listed in the NOAA tables are based on EPA methods which allow partial decomposition of sediment samples and thus reflects that portion of any element that may become biologically available/mobile under extreme environmental conditions. For example, the NOAA tables list background levels in soil/sediments for aluminum (Al) as 0.26% which reflects the average aluminum (Al) biologically available. However, our results for aluminum (Al) range from 2.85% - 7.17%, reflecting total recovery of the element by our digestions method. Aluminum (Al) is a major component of most minerals found in native rock and soils. In addition to aluminum (Al), iron (Fe) and barium (Ba) exceed the SQuirTs background levels by an order of magnitude. Concentrations of the other elements that are listed in SQuirTs are near the background levels.

Table F-4 presents a correlation matrix for elemental concentrations and textural components (Water content, bulk density, gravel, sand, silt and clay). Most elements are significantly correlated with one or more textural components, reflecting some textural control over the relative abundance of these elements. For example, many elements show a significantly strong inverse relationship with sand indicating that these elements are found in the mud (silt+clay) fraction of the sediments. Hafnium (Hf) and Uranium (U) show the least significant correlations with the other parameters measured.

Because of the wide range of sediment types analyzed, comparisons of absolute metal concentrations between the sediments are very difficult due to variation in textural characteristics. Likewise, assessing down core changes in elements is further complicated since there is a significant downcore change in the textural character of the sediments (example: Core 4, Table G-1).

To reduce the effect of grain size, metal concentrations may be discussed in terms of enrichment factors (EF). The use of enrichment factors also allows for comparisons of sediments from different environments and the comparisons of sediments whose trace metal contents were obtained by different analytical techniques (Cantillo, 1982; Hill and others, 1990; Sinex and Helz, 1981). Once metal data are "normalized" with respect to textural differences, trends in the spatial distribution of metals are easier to realize and interpret.

Enrichment factor is defined as:

$$EF_{(X)} = \frac{(X/N)_{sample}}{(X/N)_{reference}}$$

where:

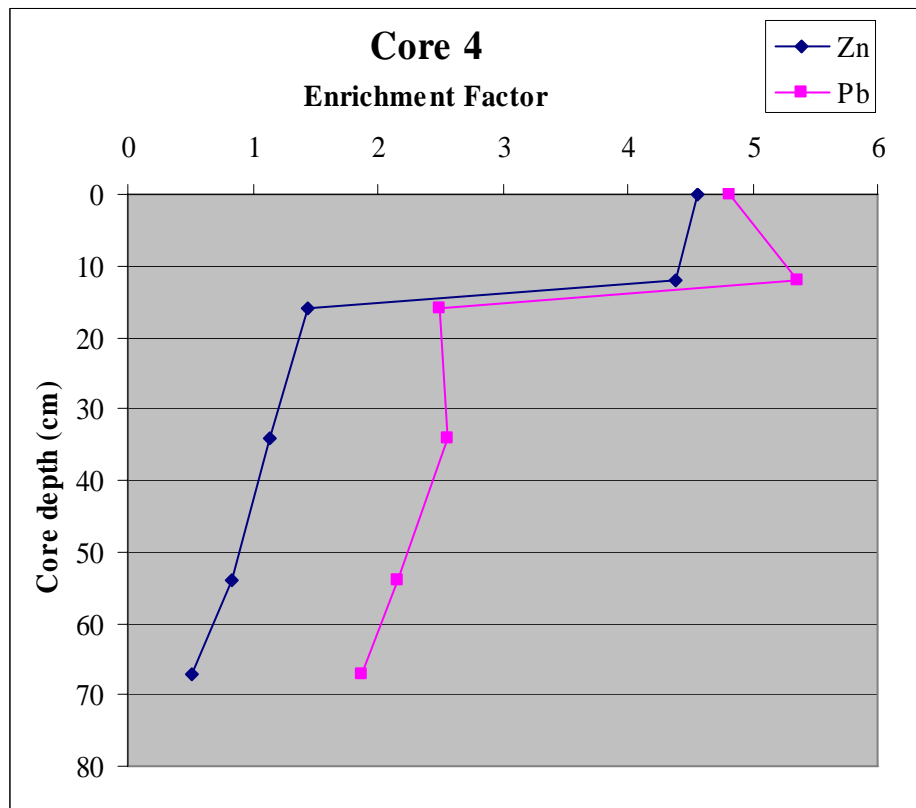
$EF_{(X)}$  is the enrichment factor for the metal X;  
 $X/N_{(sample)}$  is the ratio of the concentrations of metal X to major metal N (Fe or Al) in the sample;  
 $X/N_{(reference)}$  is the ratio of the concentrations of metal X to major metal N (Fe or Al) in a reference material, such as an average crustal rock.

Both aluminum (Al) and iron (Fe) were chosen as the element for normalizing because anthropogenic sources for these metals are small compared to natural sources (Helz, 1976). Average continental crust is used as the reference material (Taylor, 1964). Taylor's averages have been used in other studies involving various sedimentary environments, including fresh water reservoirs (Ortt et al., 1999; Sinex and Helz, 1981; Wells et al., 2007).

EF values calculated using aluminum as the normalizing element are similar to those values using iron as the normalizing elements (Table 2). EF values of one or less indicate no or under-enrichment of that element with respect to continental crust rock. In both sets of EFs, most elements show no or little enrichment (i.e.  $EF < 2$ ). However, several elements are greatly enriched with respect to the reference material. The enrichment of zinc (Zn) and lead (Pb) is due, in part, to anthropogenic sources; both of these elements have a regional atmospheric source component related to man activities. The enrichment of both elements decreases with depth identifying that concentrations of these metals have changed over time (Figure 4). Other elements having high EF values include arsenic (As), cesium (Cs), hafnium (Hf), antimony (Sb), uranium (U) and lutetium (Lu). The high enrichment may reflect a natural regional abundance of the elements as they do not exhibit any significant downcore change. For example, a source of hafnium (Hf) is zircon, a mineral found in the parent rock of the study area.

Element	Mean EF using		Element	Mean EF using	
	Fe	Al		Fe	Al
P	0.62	0.53	Sb	10.64	9.86
Cr	1.05	0.85	Rb	1.86	1.44
Cu	0.52	0.42	Sc	0.76	0.61
Fe	--	1.02	Sr	0.30	0.24
Mn	0.50	0.49	Ti	0.94	0.79
Ni	0.69	0.56	Th	1.91	1.53
Pb	3.46	2.92	U	3.10	2.43
Zn	2.62	2.04	V	0.78	0.70
Al	1.24	--	Y	1.48	1.16
As	9.13	9.23	La	2.31	1.79
Ba	2.18	1.61	Ce	1.88	1.48
Co	0.87	0.72	Nd	1.70	1.23
Cs	2.99	2.16	Sm	1.44	1.13
Eu	1.83	1.42	Yb	1.70	1.31
Hf	6.77	5.25	Lu	2.66	2.08
Mg	0.24	0.19			

**Table 2. Comparison of EF values using Fe and Al, respectively, as the normalizing metal. Values shown for each element are an average of all samples.**



**Figure 4. Plot of EF (normalized using Fe) for Zn and Pb in core 4.**

The downcore plot of lead (Pb) and zinc (Zn) in Core 4 also allows a general conclusion regarding the sediment history in that core. Both lead (Pb) and zinc (Zn) environmental levels increased dramatically in the early 1900s due to the industrial use of these metals and increased atmospheric levels due primarily to coal-powered factories and power plants and automotive exhaust. A peak in lead (Pb) concentrations during the mid-1970s is also a continental pattern observed in sediments caused by the removal of alkyl-lead from gasoline. (Owens and Cornwell, 1995) The lead (Pb) and zinc (Zn) profiles in Core 4 (Figure 4) demonstrate this trend with a significant increase in both lead (Pb) and zinc (Zn) levels between the samples collected at 16-34cm depth and 12-16cm depth. A peak in the lead (Pb) data is also displayed in the 12-16cm depth sample. While the sampling intervals used for this study are too gross for determining the exact chronology of the sediments in this core, it can be generalized that sediments deeper than 25cm [range: 16-34cm] [10 inches] are older than the early 1900s, and they are from the original Swauger's millpond. Conversely, the sediments shallower than 25 cm [range: 16-34cm] [10 inches] have been deposited since the early 1900s, and they are mostly from post-1935 dam construction.

Priority Pollutants (Semi-Volatile Organics) Analyses

Samples from cores 1, 11, and 15 were created using sediment collected along the length of the core. These samples were submitted to DHMH for EPA 8270 analyses. The report identified only pesticides from this submission (Appendix D). A second set of the same samples was submitted to DHMH for EPA 8270 analyses (Appendix E). These samples were processed using the EPA 8270 method with the exception that the holding time between collection of the samples and the analysis of the samples was exceeded. All results showed that the levels of all of the analyzed pollutants were below detection limits.

Delta Growth Analysis

Maps and aerial images from 1899, 1904, 1938, 1944, 1962, 1995, and 2007 were analyzed to document the change in the northern boundary of New Germany Lake. This area is subject to a faster sedimentation rate as it is the area where there is the greatest decrease in energy. Unfortunately, the scales, datums, and quality of the maps and images differ enough that absolute measurements are difficult to compare; however, the general trend can be observed. A point was selected where Poplar Lick Run intersects the south side of Twin Churches Road. From that point, distances were measured to the shoreline of the northern most portion of New Germany Lake. (Table 3)

Year	Document	Scale	Distance Measured (Meters)
1899	USGS Topo	1:62,500	330
1904	USGS Topo	1:62,500	342
1938	USGS Topo	1:62,500	349
1938	Imagery	1:20,000	202
1944	USGS Topo	1:24,000	207
1962	Imagery	1:20,000	211
1995	Imagery	1 Meter	238
2007	Imagery	1 Meter	238

**Table 3. Measurement of Northern Lake Boundary migration.**

The three measurements from the 1899, 1904, and 1938 topographic maps yield a southerly migration of 19 meters [62 feet] over a 39 year period. It is unknown when the actual field surveys were

performed for these maps; however, the 1938 topographic map shows the New Germany Lake at its pre-1933-1935 construction extent. These measurements provide a rough calculation of 0.5 meter/year [1.6 feet/year] growth of the headwaters delta.

The five measurements since the construction of the new dam yield a 36 meter [118 feet] southerly migration over a 69 year period. This calculates to an average 0.5 meter/year growth [1.6 feet/year] of the headwaters delta.

## **CONCLUSIONS AND RECOMMENDATIONS**

### **Sediment Accumulation**

New Germany Lake has accumulated a maximum of 33,191 cubic meters [43,412 cubic yards] of sediment within its current acreage. These sediments range from 0.56-1.16 meters [1.8- 3.8 feet] in thickness throughout the majority of the lake with the most sediment accumulating in the northern portion of the lake. Additionally, the delta formed in the headwaters of the lake has shown a history of growth at 0.5 meters per year [1.6 feet/year].

### **Sediment Physical Properties and Chemical Properties**

The accumulated sediment is generally a silty-clay. Towards the southern portion of the lake, there is an abnormally higher amount of sand in the sediment which is likely to be from the historical and current man-made beaches. Towards the northern portion of the lake, silt dominates over the clay percentages and changes the sediment classification to clayey-silt.

Chemical analyses document that the sediments are fairly consistent with average crustal standards. There is an elevation of rare earth metals; however, these metals do not reach any levels for national screening criteria. The elevation of these metals is likely due to localized geologic composition.

### **Priority Pollutants (Semi-Volatile Organics) Pollution**

No forms of sediment pollution were identified. While the second analysis of priority pollutants using EPA 8270 methods was outside of the holding period, there were no pollutants detected via the analysis. Additional research with Maryland Department of the Environment confirmed that there are no known pollutants identified within the area surrounding the park.

### **Recommendations**

No further studies are recommended to document sediment accumulation within New Germany Lake.

Should studies of this type be extended to other small-watershed lakes similar to this study area, it is recommended to take additional samples throughout the cores and analyze them for both physical characteristics and chemical composition. The additional samples would document downcore changes and assist in interpretation of the cores.

## **ACKNOWLEDGEMENTS**

The authors extend their gratitude to the following individuals for assisting in the creation of this report. Ranger Michael Gregory from New Germany State Park assisted in logistical coordination, boats, and field laboratory space. Katie Offerman of MGS performed all physical sediment analyses. Lamere Hennessee, Heather Quinn, Katie Offerman, and Dave Drummond collected, logged, documented, and

sampled all cores during the field collection. Robert Conkwright Jr. provided the geologic history of the region. Acknowledgements are also given to Robert Conkwright, Jr., Jeffrey Halka, Butch Norden, Michael Gregory, Nita Settina, Heather Quinn, and Lamere Hennessee for their review and comments on this report.

Funding for this study was provided by the Maryland Department of Natural Resources.

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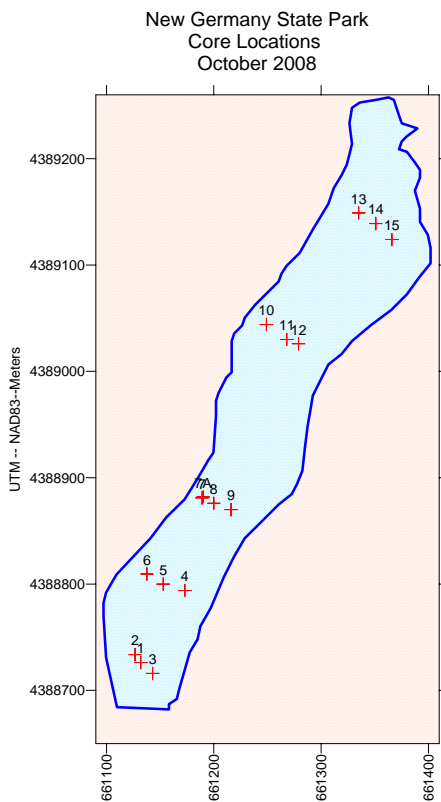


# Appendix A

## Core Locations

Core	UTM--NAD83--Meters		Water Depth FT	Water Depth Meters
	Easting	Northing		
1	661132	4388726	6.9	2.09
2	661127	4388734	6.6	2.01
3	661143	4388716	7.0	2.13
4	661173	4388794	4.9	1.49
5	661153	4388800	6.7	2.04
6	661138	4388810	5.4	1.65
7	661189	4388881	3.8	1.16
7A	661190	4388882	4.0	1.22
8	661200	4388876	5.1	1.55
9	661216	4388870	5.1	1.55
10	661249	4389044	2.5	0.76
11	661268	4389030	2.8	0.85
12	661279	4389026	2.5	0.76
13	661335	4389149	1.1	0.34
14	661351	4389139	1.4	0.43
15	661366	4389124	1.5	0.46
16	661110	4388545	0.0	0.00

**Table A-1. Locations and Water Depths of collected cores.**



**Figure A-1. Map depicting the location of the collected cores. Shoreline is approximate from 1944 USGS topographic map.**

# Appendix B

## Core Logs























































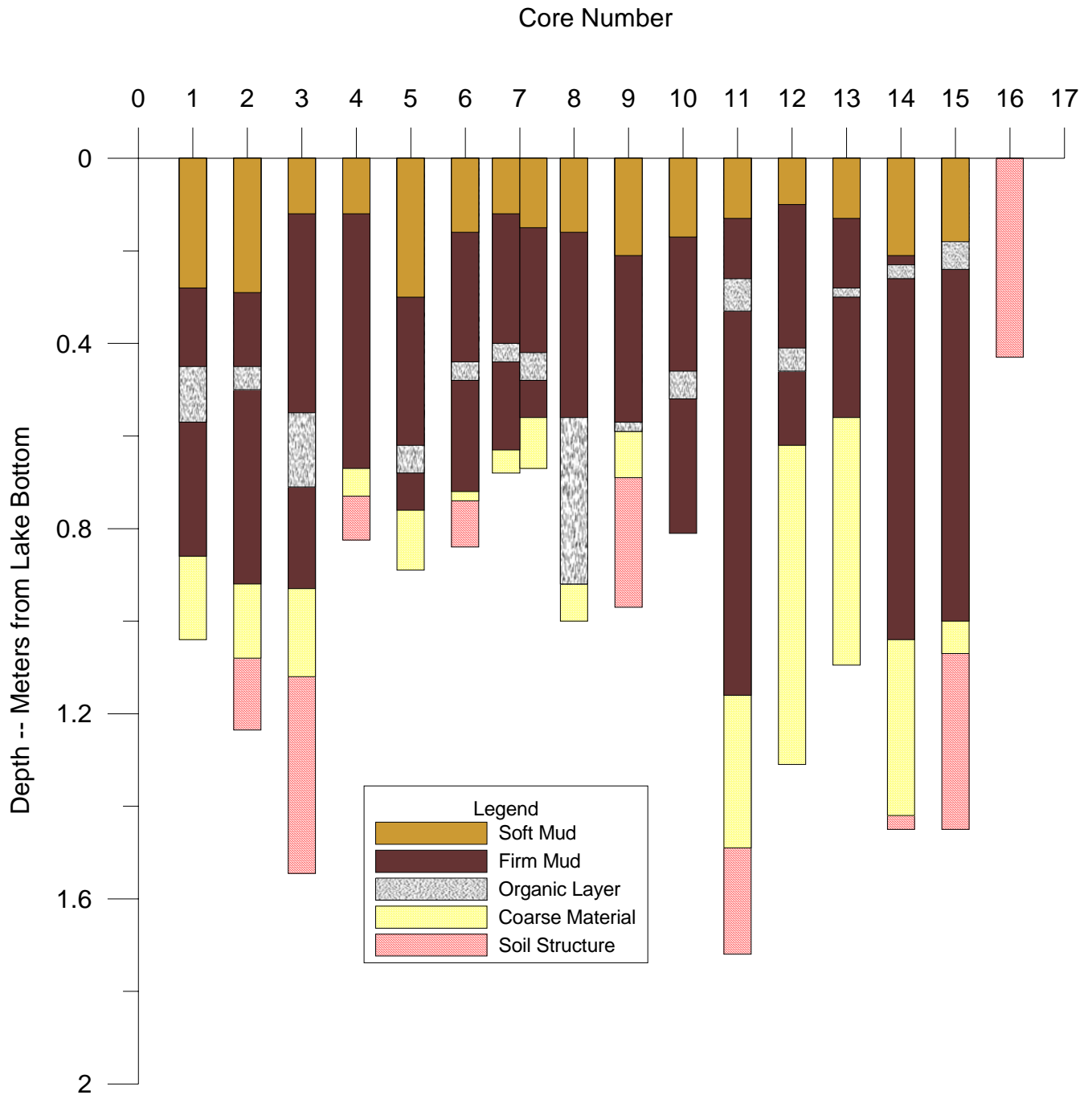
# Appendix C

## Core Stratigraphy

Core	Length	Firm Mud	Organic	End of Organic	Coarse	Soil	Easting	Northing	WaterDepth
Label	Meters	Depth-Meters	Layer Depth-Meters	Layer Depth-Meters	Material Depth-Meters	Horizon Depth-Meters	MetersUTM	NAD83	Meters
1	1.04	0.28	0.45	0.57	0.86	NO	661132	4388726	2.09
2	1.24	0.29	0.45	0.50	0.92	1.08	661127	4388734	2.01
3	1.55	0.12	0.55	0.71	0.93	1.12	661143	4388716	2.13
4	0.83	0.12	NO	NO	0.67	0.73	661173	4388794	1.49
5	0.89	0.30	0.62	0.68	0.76	NO	661153	4388800	2.04
6	0.84	0.16	0.44	0.48	0.72	0.74	661138	4388810	1.65
7	0.68	0.12	0.40	0.44	0.63	NO	661189	4388881	1.16
7A	0.67	0.15	0.42	0.48	0.56	NO	661190	4388882	1.22
8	1.00	0.16	0.56	0.92	0.92	NO	661200	4388876	1.55
9	0.97	0.21	0.57	0.59	0.59	0.69	661216	4388870	1.55
10	0.81	0.17	0.46	0.52	NO	NO	661249	4389044	0.76
11	1.72	0.13	0.26	0.33	1.16	1.49	661268	4389030	0.85
12	1.31	0.10	0.41	0.46	0.62	NO	661279	4389026	0.76
13	1.10	0.13	0.28	0.30	0.56	NO	661335	4389149	0.34
14	1.45	0.21	0.23	0.26	1.04	1.42	661351	4389139	0.43
15	1.45	0.18	0.18	0.24	1.00	1.07	661366	4389124	0.46
16	0.43	NO	NO	NO	NO	0.00	661110	4388545	0.00

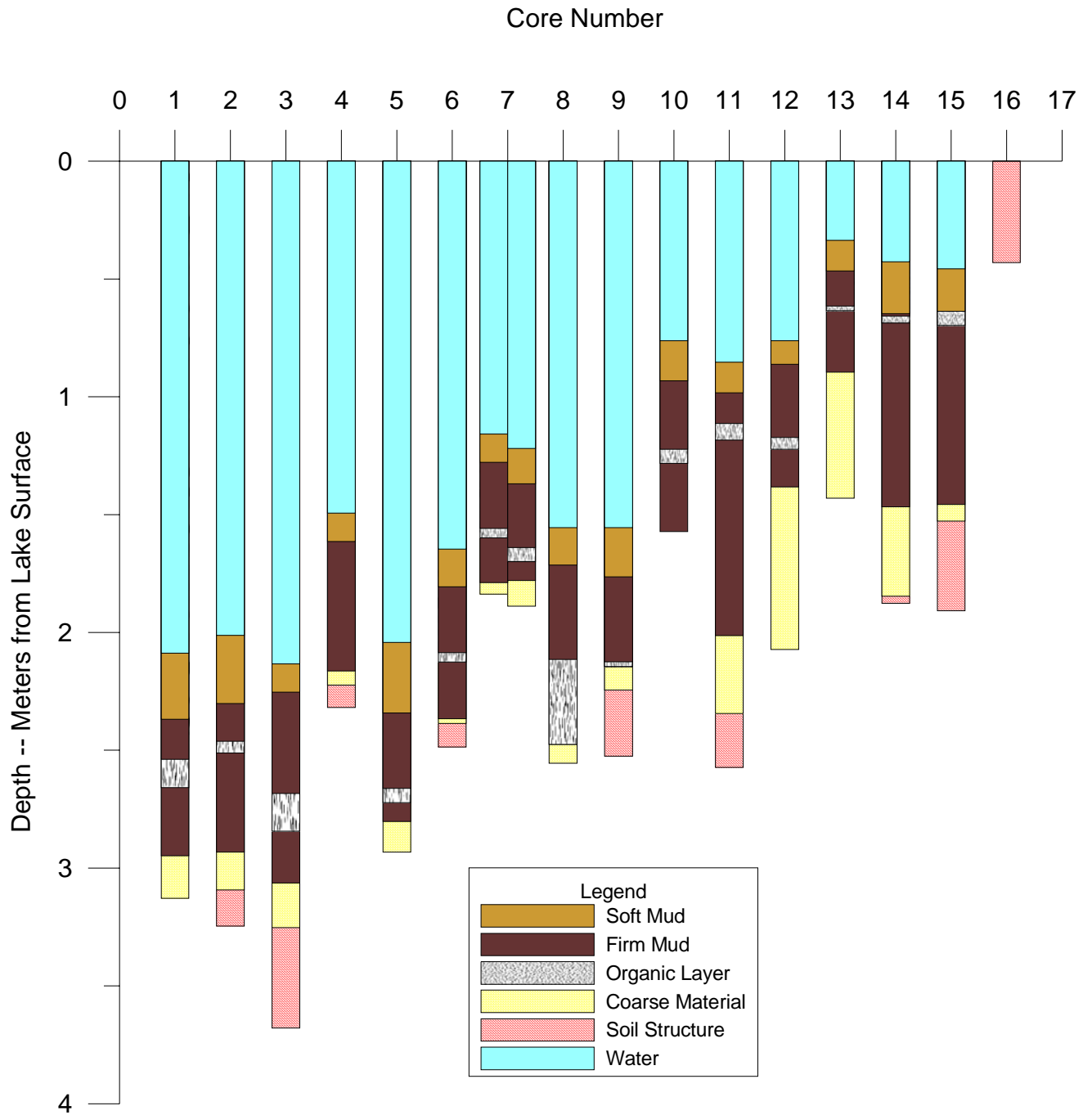
**Table C-1. Observed data recording the depth in the collected cores to the identified facies. Identification of the facies was performed through a comparison and analysis of core logs, photographs, physical properties, and elemental properties of the collected cores. NO means that the facies was not observed.**

# Identification of Selected Facies in Collected Cores New Germany State Park--October 2008



**Figure C-1.** Graph identifying selected facies versus depth in the collected cores. Depth is total sediment depth regardless of water depth at the collection location. Core locations are identified in Figure A-1. The cores are bracketed in their transects which are ordered from downstream to upstream.

# Identification of Selected Facies in Collected Cores New Germany State Park--October 2008



**Figure C-2.** Graph identifying selected facies versus depth in the collected cores. Depth is meters from lake surface.

# Appendix D

## Pesticide Analysis



Send Report To:

Richard Onth

410-554-5541

ronth@dpr.state.md.us

State of Maryland

DHMH - Laboratories Administration

Division of Environmental Chemistry

**TRACE ORGANICS SECTION**

201 W. Preston Street, Baltimore, Maryland 21201

John M. DeBoy, Dr. P.H., Director

Lab No. Date Received

990822 OCT 30 08

Do not write above this line

**LABORATORY ANALYSIS REQUEST**

Bottle No: NGSP Core 1 Plant / Site Name: NGSP County: GARRETT

Sample Source: 2300 St. Paul Street Baltimore MD 21218 Location: LAKE Sediment #1  
Street Town or City (well no., lab sink, sample tap, etc.)

Sampler ID:  PWSID:  Plant ID:

Collector: Richard Onth 410-554-5541  
(include telephone number)

Date Collected: 7 / OCT / 2008 Time Collected: 10 a.m. \_\_\_\_\_ p.m.

Field Preserved:  Yes  No Preservative Used:  1:1 HCl+Ascorbic acid  Na<sub>2</sub>SO<sub>4</sub>  6 mg NH<sub>4</sub>Cl

Sample Type:  Drinking Water  Landfill  Source (Raw Water)  Liquid  
 Community  Stream  Distribution (Treated)  Solid  
 Non-Community  Sediment  Water Treatment Plant POE  Other 8270  
 Private

Specify Program:  SDWA  NPDES  CWA  RCRA  Consumer Products  Other \_\_\_\_\_

Test Requested:  Trihalomethanes  Volatiles  Semi-volatiles  Haloacetic Acids

FIELD DATA:                 
pH Free Cl Total Cl

Field Blank Bottle No.: \_\_\_\_\_  
Trip Blank Bottle No.: \_\_\_\_\_

Remarks: Pesticides

Laboratory Supervisor: Sadia Mirza Date Reported: 11 / 19 / 08

•Phone: (410) 767-4388 •Fax: (410) 225-9318

Form Revised 12/05  
DHMH 4362

State of Maryland  
Department of Health and Mental Hygiene  
Division of Environmental Chemistry  
**TRACE ORGANICS SECTION**  
201 W. Preston Street, Baltimore, MD 21201  
*John M. Deboy, Dr. P.H., Director*

**Certificate of Analysis - Semivolatiles**

Method: 3540/8270  
Date Analyzed: 11/07/08  
Sample Name: 990822

<u>Contaminants</u>	<u>DL*</u>	<u>Results</u>
Alpha BHC	333	ND
Beta BHC	333	ND
Gamma BHC (Lindane)	333	ND
Heptachlor	333	ND
Delta BHC	333	ND
Aldrin	333	ND
Heptachlor Epoxide	333	ND
Endosulfan I	333	ND
4,4' - DDE	333	ND
Dieldrin	333	ND
4,4' - DDD	333	ND
Endosulfan II	333	ND
Endrin Aldehyde	333	ND
Endosulfan Sulfate	333	ND
4,4' - DDT	333	ND
Methoxychlor	333	ND
Endrin Ketone	333	ND

\*All results are in parts per billion

ND = Less than detection limit

e = estimated value

Section Chief: \_\_\_\_\_

*Sadia Munera*

Date Approved: \_\_\_\_\_

*11/19/08*

Phone: (410)-767-5582

Fax: (410) 225-9318

Send Report To:

State of Maryland

Lab No. Date Received

Richard O'Att

DHMH - Laboratories Administration

990823 007308

410-554-5541

Division of Environmental Chemistry

**TRACE ORGANICS SECTION**

201 W. Preston Street, Baltimore, Maryland 21201

John M. DeBoy, Dr. P.H., Director

oatt@dmr.state.md.us

Do not write above this line

**LABORATORY ANALYSIS REQUEST**

Bottle No: NGSP 11 Plant / Site Name: NGSP County: GARRET

Sample Source: 2300 St. Paul St. Baltimore MD 21218 Location: Lake Sediment # 11  
Street Town or City (well no., lab sink, sample tap, etc.)

Sampler ID:  PWSID:  Plant ID:

Collector: Richard O'Att 410-554-5541  
(include telephone number)

Date Collected: 8/10/2008 Time Collected: 10 a.m. \_\_\_\_\_ p.m.

Field Preserved:  Yes  No Preservative Used:  1:1 HCl+Ascorbic acid  Na<sub>2</sub>SO<sub>4</sub>  6 mg NH<sub>4</sub>Cl

Sample Type:  Drinking Water  Landfill  Source (Raw Water)  Liquid  
 Community  Stream  Distribution (Treated)  Solid  
 Non-Community  Sediment  Water Treatment Plant POE  Other 8270  
 Private

Specify Program:  SDWA  NPDES  CWA  RCRA  Consumer Products  Other \_\_\_\_\_

Test Requested:  Trihalomethanes  Volatiles  Semi-volatiles  Haloacetic Acids

FIELD DATA:                 
pH Free Cl Total Cl

Field Blank Bottle No.: \_\_\_\_\_  
Trip Blank Bottle No.: \_\_\_\_\_

Remarks: Pesticides

Laboratory Supervisor: Sadia Muneem

Date Reported: 11/19/08

•Phone: (410) 767-4388

•Fax: (410) 225-9318

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DHMH 4362

State of Maryland  
Department of Health and Mental Hygiene  
Division of Environmental Chemistry  
**TRACE ORGANICS SECTION**  
201 W. Preston Street, Baltimore, MD 21201  
*John M. Deboy, Dr. P.H., Director*

**Certificate of Analysis - Semivolatiles**

Method: 3540/8270  
Date Analyzed: 11/07/08  
Sample Name: 990823

<u>Contaminants</u>	<u>DL*</u>	<u>Results</u>
Alpha BHC	333	ND
Beta BHC	333	ND
Gamma BHC (Lindane)	333	ND
Heptachlor	333	ND
Delta BHC	333	ND
Aldrin	333	ND
Heptachlor Epoxide	333	ND
Endosulfan I	333	ND
4,4' - DDE	333	ND
Dieldrin	333	ND
4,4' - DDD	333	ND
Endosulfan II	333	ND
Endrin Aldehyde	333	ND
Endosulfan Sulfate	333	ND
4,4' - DDT	333	ND
Methoxychlor	333	ND
Endrin Ketone	333	ND

\*All results are in parts per billion

ND = Less than detection limit

e = estimated value

Section Chief: \_\_\_\_\_

*Sadra Klineem*

Date Approved: \_\_\_\_\_

*11/18/08*

Phone: (410)-767-5582

Fax: (410) 225-9318

Send Report To:

Richard Ott

410-554-5541

rott@dnr.state.md.us

State of Maryland

DHMH - Laboratories Administration

Division of Environmental Chemistry

**TRACE ORGANICS SECTION**

201 W. Preston Street, Baltimore, Maryland 21201

John M. DeBoy, Dr. P.H., Director

Lab No. Date Received

990824 OCT 30 8

Do not write above this line

**LABORATORY ANALYSIS REQUEST**

Bottle No: NGSP15 Plant / Site Name: NGSP County: Garnett

Sample Source: 2300 St. Paul Street Baltimore, MD <sup>21218</sup> Location: Lake Sediment # 15  
Street Town or City (well no., lab sink, sample tap, etc.)

Sampler ID:  PWSID:  Plant ID:

Collector: Richard Ott 410-554-5541  
(include telephone number)

Date Collected: 9 / OCT / 2008 Time Collected: 10 a.m. \_\_\_\_\_ p.m.

Field Preserved:  Yes  No Preservative Used:  1:1 HCl+Ascorbic acid  Na<sub>2</sub>SO<sub>4</sub>  6 mg NH<sub>4</sub>Cl

Sample Type:  Drinking Water  Landfill  Source (Raw Water)  Liquid  
 Community  Stream  Distribution (Treated)  Solid  
 Non-Community  Sediment  Water Treatment Plant POE  Other 8270  
 Private

Specify Program:  SDWA  NPDES  CWA  RCRA  Consumer Products  Other \_\_\_\_\_

Test Requested:  Trihalomethanes  Volatiles  Semi-volatiles  Haloacetic Acids

FIELD DATA:                 
pH Free Cl Total Cl

Field Blank Bottle No.: \_\_\_\_\_  
Trip Blank Bottle No.: \_\_\_\_\_

Remarks: Pesticides

Laboratory Supervisor: Sadia Muneem Date Reported: 11 / 19 / 08

•Phone: (410) 767-4388

•Fax: (410) 225-9318

Form Revised 12/05  
DHMH 4362

SUBMITTOR'S COPY

State of Maryland  
Department of Health and Mental Hygiene  
Division of Environmental Chemistry  
**TRACE ORGANICS SECTION**  
201 W. Preston Street, Baltimore, MD 21201  
*John M. Deboy, Dr. P.H., Director*

**Certificate of Analysis - Semivolatiles**

Method: 3540/8270  
Date Analyzed: 11/07/08  
Sample Name: 990824

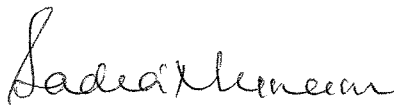
<u>Contaminants</u>	<u>DL*</u>	<u>Results</u>
Alpha BHC	333	ND
Beta BHC	333	ND
Gamma BHC (Lindane)	333	ND
Heptachlor	333	ND
Delta BHC	333	ND
Aldrin	333	ND
Heptachlor Epoxide	333	ND
Endosulfan I	333	ND
4,4' - DDE	333	ND
Dieldrin	333	ND
4,4' - DDD	333	ND
Endosulfan II	333	ND
Endrin Aldehyde	333	ND
Endosulfan Sulfate	333	ND
4,4' - DDT	333	ND
Methoxychlor	333	ND
Endrin Ketone	333	ND

\*All results are in parts per billion

ND = Less than detection limit

e = estimated value

Section Chief:



Date Approved:



Phone: (410)-767-5582

Fax: (410) 225-9318

# Appendix E

## EPA 8270 Semi-Volatile Analysis

Note: The following results are from samples which were analyzed beyond the accepted holding times established in the EPA 8270 protocol.

Send report To:  
Richard Ortt  
2300 St. Paul Street  
Baltimore, MD 21218

State of Maryland  
DHMH - Laboratories Administration  
Division of Environmental Chemistry  
**TRACE ORGANICS SECTION**  
201 W. Preston Street, Baltimore, Maryland 21201  
John M. DeBoy, Dr. P.H., Director

Lab No. Date Received  
  
991055 10/16/08  
Do not write above this line.

**LABORATORY ANALYSIS REQUEST**

Bottle No: NGSP15 Plant / Site Name: NGSP County: GARRETT

Sample Source: 2300 St. Paul Street Baltimore, MD 21218 Location: Lake Sediment #15  
Street Town or City (well no., lab sink, sample tap, etc.)

Sampler ID:   
PWSID:   
Plant ID:

Collector: Richard Ortt 410-554-5541  
(include telephone number)

Date Collected: 9 / OCT / 2008 Time Collected: 10 a.m. \_\_\_\_\_ p.m.

Field Preserved:  Yes  No Preservative Used:  1:1 HCl+Ascorbic acid  Na<sub>2</sub>SO<sub>4</sub>  6 mg NH<sub>4</sub>Cl

Sample Type:  Drinking Water  Landfill  Source (Raw Water)  Liquid  
 Community  Stream  Distribution (Treated)  Solid  
 Non-Community  Sediment  Water Treatment Plant POE  Other 8270  
 Private

Specify Program:  SDWA  NPDES  CWA  RCRA  Consumer Products  Other \_\_\_\_\_

Test Requested:  Trihalomethanes  Volatiles  Semi-volatiles  Haloacetic Acids

FIELD DATA:                 
                  pH      Free Cl      Total Cl

Field Blank Bottle No.: _____
Trip Blank Bottle No.: _____

Remarks: ANALYZE FOR FULL EPA 8270 Semi-Volatiles

Laboratory Supervisor: Richard Ortt Date Reported: 12/31/08

•Phone: (410) 767-4388 •Fax: (410) 225-9318

Form Revised 12/05  
DHMH 4362



State of Maryland  
 Department of Health and Mental Hygiene  
 Division of Environmental Chemistry  
**TRACE ORGANICS SECTION**  
 201 W. Preston Street, Baltimore, MD 21201  
 John M. Deboy, Dr. P.H., Director

## Certificate of Analysis - Semivolatiles

Method: 3540/8270  
 Date Analyzed: 12/24/08  
 Sample Name: 991065

<u>Contaminants</u>	<u>DL</u>	<u>Results</u>	<u>Contaminants</u>	<u>DL</u>	<u>Result</u>
Phenol	333	ND	4-Nitrophenol	333	ND
Bis(2-Chloroethyl)ether	333	ND	2,4-Dinitrotoluene	333	ND
2-Chlorophenol	333	ND	2,3,4,6-Tetrachlorophenol	333	ND
1,3-Dichlorobenzene	333	ND	Fluorene	333	ND
1,4-Dichlorobenzene	333	ND	Diethylphthalate	333	ND
1,2-Dichlorobenzene	333	ND	4-Chlorophenyl phenyl ether	333	ND
Bis(2-Chloroisopropyl)ether	333	ND	4,6 -Dinitro-2-methylphenol	333	ND
2-Methyl phenol	333	ND	Diphenylamine	333	ND
Hexachloroethane	333	ND	4-Bromophenyl phenyl ether	333	ND
3&4-Methyl Phenol	333	ND	1,3,5-Trinitrobenzene	333	ND
Nitrobenzene	333	ND	Phenacetin	333	ND
Isophorone	333	ND	Hexachlorobenzene	333	ND
2-Nitrophenol	333	ND	Pentachlorophenol	333	ND
2,4-Dimethylphenol	333	ND	Pentachloronitrobenzene	333	ND
Bis(2-Chloroethoxy)methane	333	ND	Phenanthrene	333	ND
2,4-Dichlorophenol	333	ND	Anthracene	333	ND
1,2,4-Trichlorobenzene	333	ND	2,4,6-Sec-butyl-Dinitrophenol	333	ND
Naphthalene	333	ND	Di-n-Butylphthalate	333	ND
2,6-Dichlorophenol	333	ND	4-Nitroquinoline-N-oxide	333	ND
Hexachloropropylene	333	ND	Methapyrilene	333	ND
Hexachlorobutadiene	333	ND	Fluoranthene	333	ND
4-Chloro-3-Methylphenol	333	ND	Pyrene	333	ND
1,2,4,5-Tetrachlorobenzene	333	ND	Di-methylaminoazobenzene	333	ND
hexachlorocyclopentadiene	333	ND	Butyl benzyl phthalate	333	ND
2,4,6-Trichlorophenol	333	ND	Benz(a)Anthracene	333	ND
2,4,5-Trichlorophenol	333	ND	Chrysene	333	ND
2-Chloronaphthalene	333	ND	Bis(2-Ethylhexyl)phthalate	333	ND
1,4-Naphthoquinone	333	ND	Di-n-octyl phthalate	333	ND
Acenaphthylene	333	ND	7,12-Dimethylbenz(a)anthracene	333	ND
Dimethylphthalate	333	ND	Benzo(b)fluoranthene	333	ND
2,6-Dinitrotoluene	333	ND	Benzo(k)fluoranthene	333	ND
Acenaphthene	333	ND	Benzo(a)pyrene	333	ND
2,4-Dinitrophenol	333	ND	Indeno(1,2,3-cd) pyrene	333	ND
Pentachlorobenzene	333	ND	Di benz(a,h)anthracene	333	ND
			Benzo(g,h,i)perylene	333	ND

\*All results are in parts per billion  
 ND = Less than detection limit  
 e = estimated value

Section Chief: Deborah Melendez

Date Approved: 12/31/08

Send Report To:

RICHARD O'H  
2300 St. Paul Street  
Baltimore, MD 21218

State of Maryland  
DHMH - Laboratories Administration  
Division of Environmental Chemistry  
**TRACE ORGANICS SECTION**  
201 W. Preston Street, Baltimore, Maryland 21201  
John M. DeBoy, Dr. P.H., Director

Lab No. Date Received

991066 DEC 16 8  
Do not write above this line

**LABORATORY ANALYSIS REQUEST**

Bottle No: NGSP 11 Plant / Site Name: NGSP County: GARRETT

Sample Source: 2300 St. Paul Street Baltimore, MD 21218 Location: Lake Sediment #11  
Street Town or City (well no., lab sink, sample tap, etc.)

Sampler ID:  PWSID:  Plant ID:

Collector: Richard O'H 410-554-5541  
(include telephone number)

Date Collected: 8/10/2008 Time Collected: 10 a.m. \_\_\_\_\_ p.m.

Field Preserved:  Yes  No Preservative Used:  1:1 HCl+Ascorbic acid  Na<sub>2</sub>SO<sub>4</sub>  6 mg NH<sub>4</sub>Cl

Sample Type:  Drinking Water  Landfill  Source (Raw Water)  Liquid  
 Community  Stream  Distribution (Treated)  Solid  
 Non-Community  Sediment  Water Treatment Plant POE  Other 8270  
 Private

Specify Program:  SDWA  NPDES  CWA  RCRA  Consumer Products  Other \_\_\_\_\_

Test Requested:  Trihalomethanes  Volatiles  Semi-volatiles  Haloacetic Acids

FIELD DATA:  pH  Free Cl  Total Cl  
Field Blank Bottle No.: \_\_\_\_\_  
Trip Blank Bottle No.: \_\_\_\_\_

Remarks: ANALYZE FOR FULL EPA 8270 Semi-volatiles

Laboratory Supervisor: Deborah Miller-John Date Reported: 12/31/08

•Phone: (410) 767-4388 •Fax: (410) 225-9318

Form Revised 12/05  
DHMH 4362

State of Maryland  
 Department of Health and Mental Hygiene  
 Division of Environmental Chemistry  
**TRACE ORGANICS SECTION**  
 201 W. Preston Street, Baltimore, MD 21201  
*John M. Deboy, Dr. P.H., Director*

## Certificate of Analysis - Semivolatiles

Method: 3510/8270  
 Date Analyzed: 12/25/08  
 Sample Name: 991066

<u>Contaminants</u>	<u>DL*</u>	<u>Results</u>	<u>Contaminants</u>	<u>DL</u>	<u>Result</u>
Phenol	333	ND	4-Nitrophenol	333	ND
Bis(2-Chloroethyl)ether	333	ND	2,4-Dinitrotoluene	333	ND
2-Chlorophenol	333	ND	2,3,4,6-Tetrachlorophenol	333	ND
1,3-Dichlorobenzene	333	ND	Fluorene	333	ND
1,4-Dichlorobenzene	333	ND	Diethylphthalate	333	ND
1,2-Dichlorobenzene	333	ND	4-Chlorophenyl phenyl ether	333	ND
Bis(2-Chloroisopropyl)ether	333	ND	4,6 -Dinitro-2-methylphenol	333	ND
2-Methyl phenol	333	ND	Diphenylamine	333	ND
Hexachloroethane	333	ND	4-Bromophenyl phenyl ether	333	ND
3&4-Methyl Phenol	333	ND	1,3,5-Trinitrobenzene	333	ND
Nitrobenzene	333	ND	Phenacetin	333	ND
Isophorone	333	ND	Hexachlorobenzene	333	ND
2-Nitrophenol	333	ND	Pentachlorophenol	333	ND
2,4-Dimethylphenol	333	ND	Pentachloronitrobenzene	333	ND
Bis(2-Chloroethoxy)methane	333	ND	Phenanthrene	333	ND
2,4-Dichlorophenol	333	ND	Anthracene	333	ND
1,2,4-Trichlorobenzene	333	ND	2,4,6-Sec-butyl-Dinitrophenol	333	ND
Naphthalene	333	ND	Di-n-Butylphthalate	333	ND
2,6-Dichlorophenol	333	ND	4-Nitroquinoline-N-oxide	333	ND
Hexachloropropylene	333	ND	Methapyrilene	333	ND
Hexachlorobutadiene	333	ND	Fluoranthene	333	ND
4-Chloro-3-Methylphenol	333	ND	Pyrene	333	ND
1,2,4,5-Tetrachlorobenzene	333	ND	Di-methylaminoazobenzene	333	ND
hexachlorocyclopentadiene	333	ND	Butyl benzyl phthalate	333	ND
2,4,6-Trichlorophenol	333	ND	Benz(a)Anthracene	333	ND
2,4,5-Trichlorophenol	333	ND	Chrysene	333	ND
2-Chloronaphthalene	333	ND	Bis(2-Ethylhexyl)phthalate	333	ND
1,4-Naphthoquinone	333	ND	Di-n-octyl phthalate	333	ND
Acenaphthylene	333	ND	7,12-Dimethylbenz(a)anthracene	333	ND
Dimethylphthalate	333	ND	Benzo(b)fluoranthene	333	ND
2,6-Dinitrotoluene	333	ND	Benzo(k)fluoranthene	333	ND
Acenaphthene	333	ND	Benzo(a)pyrene	333	ND
2,4-Dinitrophenol	333	ND	Indeno(1,2,3-cd) pyrene	333	ND
Pentachlorobenzene	333	ND	Di benz(a,h)anthracene	333	ND
			Benzo(g,h,i)perylene	333	ND

\*All results are in parts per billion  
 ND = Less than detection limit  
 e = estimated value

Section Chief: Richard A. Miller-Jack

Date Approved: 12/31/08

Send Report To:  
Richard Ort  
2300 St. Paul Street  
Baltimore, MD 21218

State of Maryland  
DHMH - Laboratories Administration  
Division of Environmental Chemistry  
**TRACE ORGANICS SECTION**  
201 W. Preston Street, Baltimore, Maryland 21201  
John M. DeBoy, Dr. P.H., Director

Lab No. Date Received  
  
Do not write above this line  
991067 DEC 16 8

**LABORATORY ANALYSIS REQUEST**

Bottle No: NGSP 601 Plant / Site Name: NGSP County: GARRETT

Sample Source: 2300 St. Paul Street Baltimore, MD 21218 Location: Lake Sediment #1  
Street Town or City (well no., lab sink, sample tap, etc.)

Sampler ID:  PWSID:  Plant ID:

Collector: RICHARD ORT 410-554-5541  
(include telephone number)

Date Collected: 7 / OCT / 2008 Time Collected: 10 a.m. \_\_\_\_\_ p.m.

Field Preserved:  Yes  No Preservative Used:  1:1 HCl+Ascorbic acid  Na<sub>2</sub>SO<sub>4</sub>  6 mg NH<sub>4</sub>Cl

Sample Type:  Drinking Water  Landfill  Source (Raw Water)  Liquid  
 Community  Stream  Distribution (Treated)  Solid  
 Non-Community  Sediment  Water Treatment Plant POE  Other 8270  
 Private

Specify Program:  SDWA  NPDES  CWA  RCRA  Consumer Products  Other \_\_\_\_\_

Test Requested:  Trihalomethanes  Volatiles  Semi-volatiles  Haloacetic Acids

FIELD DATA:  pH  Free Cl  Total Cl

Field Blank Bottle No.: _____
Trip Blank Bottle No.: _____

Remarks: ANALYZE FOR FULL EPA 8270 Semi-volatiles

Laboratory Supervisor: Richard Miller Date Reported: 12/31/08

•Phone: (410) 767-4388 •Fax: (410) 225-9318

Form Revised 12/05  
DHMH 4362

State of Maryland  
 Department of Health and Mental Hygiene  
 Division of Environmental Chemistry  
**TRACE ORGANICS SECTION**  
 201 W. Preston Street, Baltimore, MD 21201  
*John M. Deboy, Dr. P.H., Director*

## Certificate of Analysis - Semivolatiles

Method: 3540/8270  
 Date Analyzed: 12/24/08  
 Sample Name: 991067

<u>Contaminants</u>	<u>DL</u>	<u>Results</u>	<u>Contaminants</u>	<u>DL</u>	<u>Result</u>
Phenol	333	ND	4-Nitrophenol	333	ND
Bis(2-Chloroethyl)ether	333	ND	2,4-Dinitrotoluene	333	ND
2-Chlorophenol	333	ND	2,3,4,6-Tetrachlorophenol	333	ND
1,3-Dichlorobenzene	333	ND	Fluorene	333	ND
1,4-Dichlorobenzene	333	ND	Diethylphthalate	333	ND
1,2-Dichlorobenzene	333	ND	4-Chlorophenyl phenyl ether	333	ND
Bis(2-Chloroisopropyl)ether	333	ND	4,6 -Dinitro-2-methylphenol	333	ND
2-Methyl phenol	333	ND	Diphenylamine	333	ND
Hexachloroethane	333	ND	4-Bromophenyl phenyl ether	333	ND
3&4-Methyl Phenol	333	ND	1,3,5-Trinitrobenzene	333	ND
Nitrobenzene	333	ND	Phenacetin	333	ND
Isophorone	333	ND	Hexachlorobenzene	333	ND
2-Nitrophenol	333	ND	Pentachlorophenol	333	ND
2,4-Dimethylphenol	333	ND	Pentachloronitrobenzene	333	ND
Bis(2-Chloroethoxy)methane	333	ND	Phenanthrene	333	ND
2,4-Dichlorophenol	333	ND	Anthracene	333	ND
1,2,4-Trichlorobenzene	333	ND	2,4,6-Sec-butyl-Dinitrophenol	333	ND
Naphthalene	333	ND	Di-n-Butylphthalate	333	ND
2,6-Dichlorophenol	333	ND	4-Nitroquinoline-N-oxide	333	ND
Hexachloropropylene	333	ND	Methapyrilene	333	ND
Hexachlorobutadiene	333	ND	Fluoranthene	333	ND
4-Chloro-3-Methylphenol	333	ND	Pyrene	333	ND
1,2,4,5-Tetrachlorobenzene	333	ND	Di-methylaminoazobenzene	333	ND
hexachlorocyclopentadiene	333	ND	Butyl benzyl phthalate	333	ND
2,4,6-Trichlorophenol	333	ND	Benz(a)Anthracene	333	ND
2,4,5-Trichlorophenol	333	ND	Chrysene	333	ND
2-Chloronaphthalene	333	ND	Bis(2-Ethylhexyl)phthalate	333	ND
1,4-Naphthoquinone	333	ND	Di-n-octyl phthalate	333	ND
Acenaphthylene	333	ND	7,12-Dimethylbenz(a)anthracene	333	ND
Dimethylphthalate	333	ND	Benzo(b)fluoranthene	333	ND
2,6-Dinitrotoluene	333	ND	Benzo(k)fluoranthene	333	ND
Acenaphthene	333	ND	Benzo(a)pyrene	333	ND
2,4-Dinitrophenol	333	ND	Indeno(1,2,3-cd) pyrene	333	ND
Pentachlorobenzene	333	ND	Di benz(a,h)anthracene	333	ND
			Benzo(g,h,i)perylene	333	ND

\*All results are in parts per billion  
 ND = Less than detection limit  
 e = estimated value

Section Chief: Robert Miller Date Approved: 12/31/08

# Appendix F

## Elemental Analysis Data

Sample Interval	S %	P %	Cd ppm	Cr ppm	Cu ppm	Fe %	Mn ppm	Ni ppm	Pb ppm	Zn ppm	Au ppb	Ag ppm	Mo ppm	Al %	As ppm	Ba ppm	Be ppm	Bi ppm	Br ppm	Ca %	Co ppm	Cs ppm	Eu ppm	Hf ppm	Hg ppm
Detection Limit	0.01	0.001	0.3	2	1	0.01	1	1	3	1	2	0.3	1	0.01	0.5	50	1	2	0.5	0.01	1	1	0.2	1	1
Core4 0-12	0.2	0.054	1.7	81	21	4.58	388	52	49	260	<2	<0.3	1	7.17	14	570	4	<2	8	0.16	23	4	1.7	6	<1
Core4 12-16	0.11	0.055	1.1	70	21	3.7	365	46	44	202	<2	0.5	2	7.9	12	420	3	<2	<0.5	0.16	20	5	1.6	5	<1
Core4 16-34	0.01	0.022	<0.3	39	6	2.35	124	16	13	42	<2	<0.3	<1	4.01	10	200	1	<2	<0.5	0.08	6	3	0.9	7	<1
Core4 34-54	<0.01	0.011	<0.3	32	5	2.11	76	13	12	30	<2	<0.3	1	3.52	8	360	1	<2	<0.5	0.05	4	4	0.8	9	<1
Core4 54-67	<0.01	0.01	<0.3	42	8	2.72	81	15	13	28	<2	<0.3	1	3.71	16	320	1	<2	<0.5	0.04	4	3	<0.2	8	<1
Core4 67-82.5	<0.01	0.015	<0.3	30	5	4.1	166	15	17	26	<2	<0.3	1	3.68	13	270	1	<2	<0.5	0.03	6	3	0.4	6	<1
Core5 30-62	0.06	0.08	0.3	68	21	3.03	388	32	24	117	<2	0.4	1	7.12	11	700	3	<2	8	0.18	14	6	1.3	8	<1
Core5 62-76	0.02	0.011	<0.3	35	9	1.02	132	15	12	47	<2	<0.3	1	3.55	3	390	1	<2	3	0.16	6	3	0.9	12	<1
Core5 76-89	0.01	0.015	<0.3	37	14	1.22	104	13	11	34	<2	<0.3	<1	2.85	4	<50	1	<2	<0.5	0.09	6	2	0.5	9	<1
Core13 0-22	0.21	0.054	2.1	69	21	3.13	367	52	32	276	<2	<0.3	1	6.27	12	590	4	<2	10	0.17	23	5	1.3	9	<1
Core13 22-40	0.06	0.04	0.4	62	16	2.35	212	29	25	98	<2	<0.3	1	6.38	9	<50	3	<2	3	0.13	12	5	1.3	11	<1
Core16 0-13	0.01	0.05	<0.3	57	25	8.59	931	25	28	69	<2	<0.3	1	4.94	30	<50	2	<2	<0.5	0.02	16	<1	1.1	10	<1
Core16 13-43	<0.01	0.052	<0.3	41	12	4.33	416	23	19	54	<2	<0.3	2	3.67	13	<50	2	<2	<0.5	0.09	11	<1	0.9	9	<1

Sample Interval	Ir_ppb	K %	Mg %	Na %	Sb ppm	Rb ppm	Sc ppm	Se ppm	Sr ppm	Ta ppm	Ti %	Th ppm	U ppm	V ppm	W ppm	Y ppm	La ppm	Ce ppm	Nd ppm	Sm ppm	Sn %	Tb ppm	Yb ppm	Lu ppm
Detection Limit	5	0.01	0.01	0.01	0.1	15	0.1	3	1	0.5	0.01	0.2	0.5	2	1	1	0.5	3	5	0.1	0.01	0.5	0.2	0.05
Core4 0-12	<5	1.49	0.39	0.15	1.2	75	11.7	<3	79	<0.5	0.3	10.7	2.9	75	<1	24	41.4	62	23	5.5	0.01	<0.5	2.6	0.77
Core4 12-16	<5	2.09	0.44	0.14	1	139	11.8	<3	84	<0.5	0.5	12	2.9	112	<1	26	42	50	31	5	0.01	1	2.5	0.57
Core4 16-34	<5	1.15	0.19	0.08	0.9	42	5.8	<3	45	<0.5	0.1	7	3.2	21	<1	16	24.4	38	9	2.9	0.01	<0.5	1.6	0.54
Core4 34-54	<5	0.96	0.16	0.08	1.2	73	5.3	<3	39	1.7	0.2	7.2	2.6	35	<1	14	25.7	41	21	3.1	0.01	0.6	2.1	0.57
Core4 54-67	<5	1.13	0.18	0.08	1.1	57	5.8	<3	40	1.8	0.2	7.4	2.9	47	<1	14	25	46	11	2.9	0.01	<0.5	1.8	0.5
Core4 67-82.5	<5	1.29	0.17	0.07	1.7	45	6	<3	37	<0.5	0.2	7	3	65	<1	14	20.6	36	12	2.6	0.01	0.6	1.8	0.45
Core5 30-62	<5	1.95	0.44	0.22	0.8	118	11.1	<3	82	2.4	0.6	10.7	4.1	98	<1	26	42.8	69	23	5	0.01	<0.5	3	0.74
Core5 62-76	<5	0.87	0.19	0.11	0.6	66	6.3	<3	41	1	0.1	7.2	3.9	12	<1	26	30.4	48	23	3.8	0.01	<0.5	2.5	0.66
Core5 76-89	<5	0.74	0.17	0.08	0.5	<15	4.9	<3	33	1.2	0.3	6.6	3	35	<1	19	25.1	44	17	3.1	0.01	<0.5	1.7	0.46
Core13 0-22	<5	1.64	0.37	0.15	1.1	89	10.7	<3	74	1.6	0.3	10	5.5	52	<1	28	39.6	67	30	5.3	0.01	<0.5	3.3	0.78
Core13 22-40	<5	1.66	0.36	0.14	1.2	103	10.6	<3	70	<0.5	0.2	10	6.5	60	<1	27	40.4	68	27	5	0.01	1	2.8	0.61
Core16 0-13	<5	1.33	0.23	0.06	1.6	105	7.9	<3	43	<0.5	0.3	7.8	3.7	76	<1	26	26.6	53	20	3.9	0.01	<0.5	1.8	0.5
Core16 13-43	<5	1.13	0.19	0.05	1.3	<15	6.4	<3	41	<0.5	0.3	6.8	4	59	<1	24	24.2	43	<5	3.3	0.01	0.7	1.9	0.5

**Table F-1. Laboratory Results from elemental analysis of selected samples.**

Sample	Start_int	End_Int	P	Cr	Cu	Fe	Mn	Ni	Pb	Zn	As	As-corrected	Ba	Co	Cs	Eu	Hf	Mg	
Core4	0	12		0.59	0.93	0.44	0.93	0.47	0.80	4.50	4.26	8.93	7.14	1.54	1.06	1.53	1.63	2.30	0.19
Core4	12	16		0.55	0.73	0.40	0.68	0.40	0.64	3.67	3.01	6.95	5.56	1.03	0.83	1.74	1.39	1.74	0.20
Core4	16	34		0.43	0.80	0.22	0.86	0.27	0.44	2.13	1.23	11.40	9.12	0.97	0.49	2.05	1.54	4.79	0.17
Core4	34	54		0.24	0.75	0.21	0.88	0.19	0.41	2.24	1.00	10.39	8.31	1.98	0.37	3.12	1.56	7.01	0.16
Core4	54	67		0.21	0.93	0.32	1.07	0.19	0.44	2.31	0.89	19.72	15.77	1.67	0.35	2.22		5.92	0.17
Core4	67	82.5		0.32	0.67	0.20	1.63	0.39	0.45	3.04	0.83	16.15	12.92	1.42	0.54	2.24	0.75	4.47	0.16
Core5	30	62		0.88	0.79	0.44	0.62	0.47	0.49	2.22	1.93	7.06	5.65	1.90	0.65	2.31	1.25	3.08	0.22
Core5	62	76		0.24	0.81	0.38	0.42	0.32	0.46	2.23	1.56	3.86	3.09	2.13	0.56	2.32	1.74	9.27	0.19
Core5	76	89		0.41	1.07	0.74	0.63	0.32	0.50	2.54	1.40	6.42	5.13		0.69	1.93	1.20	8.66	0.21
Core13	0	22		0.68	0.91	0.50	0.73	0.51	0.91	3.36	5.18	8.75	7.00	1.82	1.21	2.19	1.42	3.94	0.21
Core13	22	40		0.49	0.80	0.38	0.54	0.29	0.50	2.58	1.81	6.45	5.16		0.62	2.15	1.40	4.73	0.20
Core16	0	13		0.79	0.95	0.76	2.54	1.63	0.56	3.73	1.64	27.77	22.21		1.07		1.53	5.55	0.16
Core16	13	43		1.11	0.92	0.49	1.72	0.98	0.69	3.41	1.73	16.20	12.96		0.99		1.68	6.73	0.18
			Mean	0.53	0.85	0.42	1.02	0.49	0.56	2.92	2.04	11.54	9.23	1.61	0.72	2.16	1.42	5.25	0.19

Sample	Start_int	End_Int	Sb	Rb	Sc	Sr	Ti	Th	U	V	Y	La	Ce	Nd	Sm	Yb	Lu	
Core4	0	12	6.89	0.96	0.61	0.24	0.62	1.28	1.23	0.64	0.83	1.58	1.19	0.94	1.05	0.99	1.77	
Core4	12	16	5.21	1.61	0.56	0.23	0.95	1.30	1.12	0.86	0.82	1.46	0.87	1.15	0.87	0.87	1.19	
Core4	16	34	9.24	0.96	0.54	0.25	0.40	1.50	2.43	0.32	1.00	1.67	1.30	0.66	0.99	1.09	2.22	
Core4	34	54	14.03	1.90	0.56	0.24	0.82	1.75	2.25	0.61	0.99	2.00	1.60	1.75	1.21	1.64	2.67	
Core4	54	67	12.20	1.40	0.58	0.24	0.78	1.71	2.38	0.77	0.94	1.85	1.70	0.87	1.07	1.33	2.22	
Core4	67	82.5	19.01	1.12	0.61	0.22	0.82	1.63	2.48	1.08	0.95	1.54	1.34	0.96	0.97	1.34	2.01	
Core5	30	62	4.62	1.52	0.58	0.25	1.16	1.29	1.76	0.84	0.91	1.65	1.33	0.95	0.96	1.16	1.71	
Core5	62	76	6.95	1.70	0.66	0.25	0.28	1.74	3.35	0.21	1.83	2.35	1.85	1.90	1.47	1.93	3.06	
Core5	76	89	7.22		0.64	0.25	1.27	1.99	3.21	0.75	1.66	2.42	2.12	1.75	1.49	1.64	2.66	
Core13	0	22	7.22	1.30	0.64	0.26	0.76	1.37	2.67	0.51	1.11	1.73	1.47	1.41	1.16	1.44	2.05	
Core13	22	40	7.74	1.48	0.62	0.24	0.54	1.34	3.11	0.57	1.06	1.74	1.46	1.24	1.07	1.20	1.57	
Core16	0	13	13.33	1.94	0.60	0.19	0.82	1.35	2.28	0.94	1.31	1.48	1.47	1.19	1.08	1.00	1.67	
Core16	13	43	14.58		0.65	0.25	0.98	1.59	3.32	0.98	1.63	1.81	1.61		1.23	1.42	2.24	
			Mean	9.86	1.44	0.61	0.24	0.79	1.53	2.43	0.70	1.16	1.79	1.48	1.23	1.13	1.31	2.08

**Table F-2. Enrichment Factors based on Aluminum**



Sample	Start_int	End_Int	P	Cr	Cu	Mn	Ni	Pb	Zn	Al	As	As-correct	Ba	Co	Cs	Eu	Hf	Mg	
Core4	0	12	12	0.63	1.00	0.47	0.50	0.85	4.82	4.57	1.07	9.56	7.65	1.65	1.13	1.64	1.74	2.46	0.21
Core4	12	16	16	0.80	1.07	0.58	0.58	0.93	5.36	4.39	1.46	10.14	8.12	1.50	1.22	2.54	2.03	2.54	0.29
Core4	16	34	34	0.50	0.93	0.26	0.31	0.51	2.49	1.44	1.17	13.31	10.65	1.13	0.57	2.40	1.80	5.59	0.20
Core4	34	54	54	0.28	0.85	0.24	0.21	0.46	2.56	1.14	1.14	11.86	9.49	2.26	0.43	3.56	1.78	8.00	0.18
Core4	54	67	67	0.20	0.87	0.30	0.18	0.41	2.15	0.83	0.93	18.40	14.72	1.56	0.33	2.07		5.52	0.16
Core4	67	82.5	82.5	0.20	0.41	0.12	0.24	0.27	1.87	0.51	0.61	9.92	7.93	0.87	0.33	1.37	0.46	2.75	0.10
Core5	30	62	62	1.42	1.26	0.71	0.76	0.79	3.57	3.11	1.61	11.35	9.08	3.06	1.04	3.72	2.01	4.95	0.35
Core5	62	76	76	0.58	1.93	0.90	0.77	1.10	5.30	3.71	2.38	9.20	7.36	5.07	1.32	5.52	4.14	22.08	0.45
Core5	76	89	89	0.66	1.71	1.17	0.51	0.80	4.06	2.24	1.60	10.26	8.20		1.11	3.08	1.92	13.84	0.34
Core13	0	22	22	0.93	1.24	0.69	0.69	1.25	4.60	7.09	1.37	11.99	9.59	2.50	1.65	3.00	1.95	5.40	0.29
Core13	22	40	40	0.91	1.49	0.70	0.53	0.93	4.79	3.35	1.86	11.98	9.58		1.15	3.99	2.60	8.78	0.37
Core16	0	13	13	0.31	0.37	0.30	0.64	0.22	1.47	0.65	0.39	10.92	8.74		0.42		0.60	2.18	0.06
Core16	13	43	43	0.64	0.53	0.28	0.57	0.40	1.98	1.00	0.58	9.39	7.51		0.57		0.98	3.90	0.11
		Mean		0.62	1.05	0.52	0.50	0.69	3.46	2.62	1.24	11.41	9.13	2.18	0.87	2.99	1.83	6.77	0.24

Sample	Start_int	End_Int	Sb	Rb	Sc	Sr	Ti	Th	U	V	Y	La	Ce	Nd	Sm	Yb	Lu	
Core4	0	12	12	7.38	1.02	0.65	0.26	0.67	1.37	1.32	0.68	0.89	1.70	1.27	1.01	1.13	1.07	1.89
Core4	12	16	16	7.61	2.35	0.82	0.34	1.39	1.90	1.63	1.26	1.20	2.13	1.27	1.68	1.27	1.27	1.73
Core4	16	34	34	10.78	1.12	0.63	0.29	0.46	1.75	2.84	0.37	1.16	1.95	1.52	0.77	1.16	1.28	2.59
Core4	34	54	54	16.01	2.16	0.64	0.28	0.94	2.00	2.57	0.69	1.13	2.29	1.82	2.00	1.38	1.87	3.04
Core4	54	67	67	11.38	1.31	0.55	0.22	0.73	1.60	2.22	0.72	0.88	1.72	1.59	0.81	1.00	1.24	2.07
Core4	67	82.5	82.5	11.67	0.69	0.37	0.14	0.51	1.00	1.53	0.66	0.58	0.94	0.82	0.59	0.60	0.82	1.24
Core5	30	62	62	7.43	2.44	0.94	0.41	1.86	2.07	2.82	1.35	1.46	2.65	2.14	1.53	1.55	1.86	2.75
Core5	62	76	76	16.56	4.05	1.58	0.60	0.68	4.14	7.97	0.49	4.35	5.59	4.42	4.53	3.50	4.60	7.29
Core5	76	89	89	11.54		1.03	0.41	2.02	3.17	5.13	1.20	2.66	3.86	3.38	2.80	2.38	2.62	4.25
Core13	0	22	22	9.89	1.78	0.87	0.35	1.04	1.87	3.66	0.69	1.53	2.37	2.01	1.93	1.59	1.98	2.81
Core13	22	40	40	14.37	2.74	1.15	0.45	1.01	2.50	5.77	1.06	1.96	3.23	2.72	2.31	2.00	2.24	2.92
Core16	0	13	13	5.24	0.76	0.24	0.08	0.32	0.53	0.90	0.37	0.52	0.58	0.58	0.47	0.43	0.39	0.66
Core16	13	43	43	8.45		0.38	0.14	0.57	0.92	1.93	0.57	0.95	1.05	0.93		0.72	0.82	1.30
		Mean		10.64	1.86	0.76	0.30	0.94	1.91	3.10	0.78	1.48	2.31	1.88	1.70	1.44	1.70	2.66

**Table F-3. Enrichment Factors based on Iron.**

Table F-4 . Correlation matrix for element concentrations and sediment textural data based on selected core samples. The correlations were performed using Pearson product-moment technique. Values listed in table are Pearson correlation coefficients (r). Sample sizes (N) for correlations range from 9 to 13. Critical value (Student's t distribution with N-2 degrees of freedom) at 95% is 2.2. Non significant r-values are indicated by grey type.

	% H2O	Blk D.	GRAVEL	SAND	SILT	CLAY	Mg	Sb	Rb	Sc	Sr	Ti	Th	U	V	Y	La	Ce	Nd	Sm	Yb	Lu	
% H2O	1.00																						
Blk D.	-0.98	1.00																					
GRAVEL	-0.65	0.77	1.00																				
SAND	-0.77	0.67	0.09	1.00																			
SILT	0.95	-0.94	-0.70	-0.68	1.00																		
CLAY	0.88	-0.88	-0.62	-0.77	0.75	1.00																	
Mg	0.83	-0.78	-0.31	-0.93	0.72	0.89	1.00																
Sb	-0.35	0.46	0.63	-0.11	-0.33	-0.28	-0.08	1.00															
Rb	0.53	-0.50	-0.14	-0.64	0.47	0.54	0.78	-0.22	1.00														
Sc	0.81	-0.74	-0.21	-0.96	0.69	0.84	0.98	0.06	0.73	1.00													
Sr	0.86	-0.81	-0.37	-0.92	0.75	0.92	0.99	-0.08	0.73	0.97	1.00												
Ti	0.45	-0.40	-0.07	-0.69	0.34	0.64	0.81	-0.02	0.83	0.73	0.76	1.00											
Th	0.81	-0.77	-0.35	-0.88	0.69	0.89	0.98	-0.03	0.77	0.97	0.98	0.78	1.00										
U	0.42	-0.37	-0.02	-0.50	0.58	0.15	0.39	0.02	0.37	0.43	0.37	0.07	0.31	1.00									
V	0.28	-0.20	0.18	-0.67	0.13	0.52	0.74	0.33	0.68	0.73	0.69	0.90	0.75	0.02	1.00								
Y	0.58	-0.50	0.11	-0.77	0.49	0.42	0.69	-0.14	0.71	0.73	0.64	0.46	0.62	0.64	0.41	1.00							
La	0.91	-0.87	-0.44	-0.89	0.82	0.89	0.96	-0.22	0.77	0.95	0.96	0.69	0.95	0.46	0.57	0.73	1.00						
Ce	0.79	-0.72	-0.22	-0.86	0.78	0.65	0.82	-0.10	0.64	0.83	0.80	0.56	0.76	0.68	0.44	0.76	0.87	1.00					
Nd	0.67	-0.61	-0.16	-0.70	0.62	0.52	0.73	-0.11	0.84	0.75	0.71	0.55	0.74	0.50	0.46	0.82	0.80	0.68	1.00				
Sm	0.88	-0.80	-0.27	-0.93	0.77	0.80	0.93	-0.09	0.73	0.96	0.93	0.63	0.91	0.50	0.56	0.82	0.97	0.90	0.83	1.00			
Yb	0.88	-0.83	-0.42	-0.82	0.90	0.72	0.81	-0.18	0.64	0.80	0.82	0.52	0.77	0.62	0.35	0.71	0.88	0.86	0.81	0.87	1.00		
Lu	0.89	-0.84	-0.46	-0.78	0.84	0.76	0.71	-0.25	0.38	0.72	0.75	0.37	0.67	0.39	0.18	0.59	0.81	0.79	0.61	0.82	0.89	1.00	

	<b>%H2O</b>	<b>Blk D.</b>	<b>GRAVEL</b>	<b>SAND</b>	<b>SILT</b>	<b>CLAY</b>	<b>P</b>	<b>Cr</b>	<b>Cu</b>	<b>Fe</b>	<b>Mn</b>	<b>Ni</b>	<b>Pb</b>	<b>Zn</b>	<b>Al</b>	<b>As</b>	<b>Ba</b>	<b>Co</b>	<b>Cs</b>	<b>Eu</b>	<b>Hf</b>	
<b>%H2O</b>	<b>1.00</b>																					
<b>Blk D.</b>	<b>-0.98</b>	<b>1.00</b>																				
<b>GRAVEL</b>	<b>-0.65</b>	<b>0.77</b>	<b>1.00</b>																			
<b>SAND</b>	<b>-0.77</b>	<b>0.67</b>	<b>0.09</b>	<b>1.00</b>																		
<b>SILT</b>	<b>0.95</b>	<b>-0.94</b>	<b>-0.70</b>	<b>-0.68</b>	<b>1.00</b>																	
<b>CLAY</b>	<b>0.88</b>	<b>-0.88</b>	<b>-0.62</b>	<b>-0.77</b>	<b>0.75</b>	<b>1.00</b>																
<b>P</b>	<b>0.49</b>	<b>-0.38</b>	<b>0.18</b>	<b>-0.88</b>	<b>0.36</b>	<b>0.59</b>	<b>1.00</b>															
<b>Cr</b>	<b>0.77</b>	<b>-0.69</b>	<b>-0.14</b>	<b>-0.92</b>	<b>0.62</b>	<b>0.78</b>	<b>0.82</b>	<b>1.00</b>														
<b>Cu</b>	<b>0.51</b>	<b>-0.41</b>	<b>0.18</b>	<b>-0.79</b>	<b>0.37</b>	<b>0.47</b>	<b>0.82</b>	<b>0.87</b>	<b>1.00</b>													
<b>Fe</b>	<b>-0.25</b>	<b>0.38</b>	<b>0.80</b>	<b>-0.32</b>	<b>-0.38</b>	<b>-0.16</b>	<b>0.46</b>	<b>0.34</b>	<b>0.54</b>	<b>1.00</b>												
<b>Mn</b>	<b>-0.03</b>	<b>0.16</b>	<b>0.72</b>	<b>-0.51</b>	<b>-0.16</b>	<b>-0.01</b>	<b>0.67</b>	<b>0.51</b>	<b>0.77</b>	<b>0.90</b>	<b>1.00</b>											
<b>Ni</b>	<b>0.79</b>	<b>-0.68</b>	<b>-0.16</b>	<b>-0.89</b>	<b>0.64</b>	<b>0.73</b>	<b>0.75</b>	<b>0.93</b>	<b>0.76</b>	<b>0.29</b>	<b>0.43</b>	<b>1.00</b>										
<b>Pb</b>	<b>0.63</b>	<b>-0.53</b>	<b>-0.02</b>	<b>-0.81</b>	<b>0.42</b>	<b>0.68</b>	<b>0.70</b>	<b>0.90</b>	<b>0.76</b>	<b>0.46</b>	<b>0.52</b>	<b>0.93</b>	<b>1.00</b>									
<b>Zn</b>	<b>0.84</b>	<b>-0.74</b>	<b>-0.27</b>	<b>-0.84</b>	<b>0.70</b>	<b>0.73</b>	<b>0.65</b>	<b>0.88</b>	<b>0.70</b>	<b>0.18</b>	<b>0.33</b>	<b>0.98</b>	<b>0.88</b>	<b>1.00</b>								
<b>Al</b>	<b>0.77</b>	<b>-0.71</b>	<b>-0.23</b>	<b>-0.92</b>	<b>0.64</b>	<b>0.86</b>	<b>0.80</b>	<b>0.94</b>	<b>0.75</b>	<b>0.26</b>	<b>0.40</b>	<b>0.88</b>	<b>0.87</b>	<b>0.82</b>	<b>1.00</b>							
<b>As</b>	<b>-0.29</b>	<b>0.40</b>	<b>0.73</b>	<b>-0.22</b>	<b>-0.37</b>	<b>-0.21</b>	<b>0.35</b>	<b>0.30</b>	<b>0.46</b>	<b>0.95</b>	<b>0.82</b>	<b>0.20</b>	<b>0.35</b>	<b>0.10</b>	<b>0.20</b>	<b>1.00</b>						
<b>Ba</b>	<b>0.45</b>	<b>-0.42</b>	<b>-0.12</b>	<b>-0.51</b>	<b>0.35</b>	<b>0.46</b>	<b>0.61</b>	<b>0.55</b>	<b>0.65</b>	<b>0.02</b>	<b>0.29</b>	<b>0.40</b>	<b>0.36</b>	<b>0.38</b>	<b>0.44</b>	<b>-0.08</b>	<b>1.00</b>					
<b>Co</b>	<b>0.68</b>	<b>-0.55</b>	<b>0.04</b>	<b>-0.89</b>	<b>0.52</b>	<b>0.62</b>	<b>0.80</b>	<b>0.92</b>	<b>0.87</b>	<b>0.47</b>	<b>0.62</b>	<b>0.97</b>	<b>0.93</b>	<b>0.94</b>	<b>0.84</b>	<b>0.36</b>	<b>0.24</b>	<b>1.00</b>				
<b>Cs</b>	<b>0.58</b>	<b>-0.52</b>	<b>-0.09</b>	<b>-0.84</b>	<b>0.57</b>	<b>0.65</b>	<b>0.82</b>	<b>0.73</b>	<b>0.60</b>	<b>0.21</b>	<b>0.39</b>	<b>0.67</b>	<b>0.57</b>	<b>0.59</b>	<b>0.83</b>	<b>0.17</b>	<b>0.41</b>	<b>0.64</b>	<b>1.00</b>			
<b>Eu</b>	<b>0.42</b>	<b>-0.35</b>	<b>-0.08</b>	<b>-0.50</b>	<b>0.28</b>	<b>0.48</b>	<b>0.44</b>	<b>0.55</b>	<b>0.47</b>	<b>0.22</b>	<b>0.22</b>	<b>0.67</b>	<b>0.73</b>	<b>0.66</b>	<b>0.54</b>	<b>0.02</b>	<b>0.45</b>	<b>0.67</b>	<b>0.23</b>	<b>1.00</b>		
<b>Hf</b>	<b>-0.10</b>	<b>0.09</b>	<b>0.15</b>	<b>0.17</b>	<b>0.06</b>	<b>-0.42</b>	<b>-0.19</b>	<b>-0.25</b>	<b>-0.03</b>	<b>-0.18</b>	<b>0.04</b>	<b>-0.34</b>	<b>-0.46</b>	<b>-0.32</b>	<b>-0.35</b>	<b>-0.15</b>	<b>-0.30</b>	<b>-0.27</b>	<b>-0.07</b>	<b>-0.58</b>	<b>1.00</b>	
<b>Mg</b>	<b>0.83</b>	<b>-0.78</b>	<b>-0.31</b>	<b>-0.93</b>	<b>0.72</b>	<b>0.89</b>	<b>0.82</b>	<b>0.93</b>	<b>0.76</b>	<b>0.13</b>	<b>0.33</b>	<b>0.87</b>	<b>0.81</b>	<b>0.82</b>	<b>0.98</b>	<b>0.07</b>	<b>0.85</b>	<b>0.82</b>	<b>0.84</b>	<b>0.53</b>	<b>-0.29</b>	
<b>Sb</b>	<b>-0.35</b>	<b>0.46</b>	<b>0.63</b>	<b>-0.11</b>	<b>-0.33</b>	<b>-0.28</b>	<b>0.11</b>	<b>0.03</b>	<b>0.03</b>	<b>0.73</b>	<b>0.45</b>	<b>0.09</b>	<b>0.23</b>	<b>0.01</b>	<b>0.06</b>	<b>0.71</b>		<b>0.17</b>	<b>0.14</b>	<b>0.17</b>	<b>-0.22</b>	
<b>Rb</b>	<b>0.53</b>	<b>-0.50</b>	<b>-0.14</b>	<b>-0.64</b>	<b>0.47</b>	<b>0.54</b>	<b>0.70</b>	<b>0.67</b>	<b>0.73</b>	<b>0.06</b>	<b>0.34</b>	<b>0.61</b>	<b>0.56</b>	<b>0.55</b>	<b>0.72</b>	<b>-0.02</b>	<b>0.62</b>	<b>0.62</b>	<b>0.74</b>	<b>0.41</b>	<b>-0.03</b>	
<b>Sc</b>	<b>0.81</b>	<b>-0.74</b>	<b>-0.21</b>	<b>-0.96</b>	<b>0.69</b>	<b>0.84</b>	<b>0.82</b>	<b>0.96</b>	<b>0.79</b>	<b>0.25</b>	<b>0.42</b>	<b>0.91</b>	<b>0.88</b>	<b>0.86</b>	<b>0.99</b>	<b>0.18</b>	<b>0.66</b>	<b>0.88</b>	<b>0.82</b>	<b>0.56</b>	<b>-0.25</b>	
<b>Sr</b>	<b>0.86</b>	<b>-0.81</b>	<b>-0.37</b>	<b>-0.92</b>	<b>0.75</b>	<b>0.92</b>	<b>0.79</b>	<b>0.92</b>	<b>0.69</b>	<b>0.09</b>	<b>0.27</b>	<b>0.88</b>	<b>0.82</b>	<b>0.84</b>	<b>0.98</b>	<b>0.03</b>	<b>0.87</b>	<b>0.81</b>	<b>0.83</b>	<b>0.53</b>	<b>-0.32</b>	
<b>Ti</b>	<b>0.45</b>	<b>-0.40</b>	<b>-0.07</b>	<b>-0.69</b>	<b>0.34</b>	<b>0.64</b>	<b>0.82</b>	<b>0.70</b>	<b>0.70</b>	<b>0.26</b>	<b>0.41</b>	<b>0.63</b>	<b>0.62</b>	<b>0.55</b>	<b>0.77</b>	<b>0.20</b>	<b>0.68</b>	<b>0.63</b>	<b>0.75</b>	<b>0.47</b>	<b>-0.43</b>	
<b>Th</b>	<b>0.81</b>	<b>-0.77</b>	<b>-0.35</b>	<b>-0.88</b>	<b>0.69</b>	<b>0.89</b>	<b>0.74</b>	<b>0.91</b>	<b>0.71</b>	<b>0.13</b>	<b>0.29</b>	<b>0.87</b>	<b>0.85</b>	<b>0.82</b>	<b>0.99</b>	<b>0.08</b>	<b>0.83</b>	<b>0.81</b>	<b>0.81</b>	<b>0.54</b>	<b>-0.35</b>	
<b>U</b>	<b>0.42</b>	<b>-0.37</b>	<b>-0.02</b>	<b>-0.50</b>	<b>0.58</b>	<b>0.15</b>	<b>0.37</b>	<b>0.35</b>	<b>0.34</b>	<b>-0.05</b>	<b>0.17</b>	<b>0.31</b>	<b>0.11</b>	<b>0.28</b>	<b>0.33</b>	<b>-0.05</b>	<b>0.07</b>	<b>0.32</b>	<b>0.51</b>	<b>0.00</b>	<b>0.55</b>	
<b>V</b>	<b>0.28</b>	<b>-0.20</b>	<b>0.18</b>	<b>-0.67</b>	<b>0.13</b>	<b>0.52</b>	<b>0.77</b>	<b>0.69</b>	<b>0.68</b>	<b>0.53</b>	<b>0.56</b>	<b>0.61</b>	<b>0.73</b>	<b>0.49</b>	<b>0.77</b>	<b>0.46</b>	<b>0.21</b>	<b>0.65</b>	<b>0.68</b>	<b>0.53</b>	<b>-0.52</b>	
<b>Y</b>	<b>0.58</b>	<b>-0.50</b>	<b>0.11</b>	<b>-0.77</b>	<b>0.49</b>	<b>0.42</b>	<b>0.73</b>	<b>0.70</b>	<b>0.80</b>	<b>0.24</b>	<b>0.58</b>	<b>0.67</b>	<b>0.58</b>	<b>0.62</b>	<b>0.65</b>	<b>0.12</b>	<b>0.16</b>	<b>0.74</b>	<b>0.62</b>	<b>0.28</b>	<b>0.32</b>	
<b>La</b>	<b>0.91</b>	<b>-0.87</b>	<b>-0.44</b>	<b>-0.89</b>	<b>0.82</b>	<b>0.89</b>	<b>0.72</b>	<b>0.90</b>	<b>0.71</b>	<b>-0.03</b>	<b>0.22</b>	<b>0.83</b>	<b>0.75</b>	<b>0.81</b>	<b>0.93</b>	<b>-0.08</b>		<b>0.76</b>	<b>0.80</b>	<b>0.43</b>	<b>-0.10</b>	
<b>Ce</b>	<b>0.79</b>	<b>-0.72</b>	<b>-0.22</b>	<b>-0.86</b>	<b>0.78</b>	<b>0.65</b>	<b>0.72</b>	<b>0.84</b>	<b>0.75</b>	<b>0.11</b>	<b>0.36</b>	<b>0.71</b>	<b>0.57</b>	<b>0.68</b>	<b>0.77</b>	<b>0.10</b>	<b>0.76</b>	<b>0.70</b>	<b>0.76</b>	<b>0.23</b>	<b>0.20</b>	
<b>Nd</b>	<b>0.67</b>	<b>-0.61</b>	<b>-0.16</b>	<b>-0.70</b>	<b>0.62</b>	<b>0.52</b>	<b>0.59</b>	<b>0.67</b>	<b>0.66</b>	<b>0.04</b>	<b>0.33</b>	<b>0.72</b>	<b>0.63</b>	<b>0.71</b>	<b>0.70</b>	<b>-0.08</b>	<b>0.50</b>	<b>0.72</b>	<b>0.71</b>	<b>0.42</b>	<b>0.16</b>	
<b>Sm</b>	<b>0.88</b>	<b>-0.80</b>	<b>-0.27</b>	<b>-0.93</b>	<b>0.77</b>	<b>0.80</b>	<b>0.76</b>	<b>0.94</b>	<b>0.81</b>	<b>0.15</b>	<b>0.39</b>	<b>0.90</b>	<b>0.82</b>	<b>0.88</b>	<b>0.91</b>	<b>0.07</b>	<b>0.76</b>	<b>0.88</b>	<b>0.77</b>	<b>0.49</b>	<b>-0.04</b>	
<b>Yb</b>	<b>0.88</b>	<b>-0.83</b>	<b>-0.42</b>	<b>-0.82</b>	<b>0.90</b>	<b>0.72</b>	<b>0.60</b>	<b>0.71</b>	<b>0.54</b>	<b>-0.14</b>	<b>0.12</b>	<b>0.73</b>	<b>0.53</b>	<b>0.75</b>	<b>0.74</b>	<b>-0.19</b>	<b>0.95</b>	<b>0.65</b>	<b>0.79</b>	<b>0.25</b>	<b>0.13</b>	
<b>Lu</b>	<b>0.89</b>	<b>-0.84</b>	<b>-0.46</b>	<b>-0.78</b>	<b>0.84</b>	<b>0.76</b>	<b>0.55</b>	<b>0.71</b>	<b>0.48</b>	<b>-0.12</b>	<b>0.11</b>	<b>0.72</b>	<b>0.54</b>	<b>0.77</b>	<b>0.66</b>	<b>-0.16</b>	<b>0.94</b>	<b>0.65</b>	<b>0.63</b>	<b>0.21</b>	<b>0.05</b>	

Element	Symbol	Units	Buffalo River NIST SRM 8702					Inorganics in Marine Sed-NIST SRM 2702					PACS-2							
			Certified/Reference Value		MGS Results			Certified/Reference Value		MGS Results			Certified/Reference Values			MGS Results				
			Certified	Std	Ave.	Std	% Recovery	Certified	Std	Ave.	Std	% Recovery	Certified	Std	Ave.	Std	% Recovery			
Sulfur	S	%				0.355	0.010				1.5		1.5375	0.043	102.50	1.29	0.13	1.2425	0.062	96.32
Phosphorus	P	%				0.0905	0.003				0.1552	0.006	0.14125	0.003	91.01	0.096	0.004	0.09425	0.004	98.18
Cadmium	Cd	ppm	2.94	0.29		3	0.115	102.04			0.817	0.011	0.95	0.058	116.28	2.11	0.15	2.225	0.150	105.45
Chromium	Cr	ppm	121.9	3.8		120	5.888	98.44			352	22	332.5	20.158	94.46	90.7	4.6	91.5	3.697	100.88
Copper	Cu	ppm				84.5	2.646				117.7	5.6	110.5	4.933	93.88	310	12	303.5	15.674	97.90
Iron	Fe	%	3.97	0.1		3.9575	0.100	99.69			7.91	0.24	7.3775	0.176	93.27	4.09	0.06	4.0425	0.158	98.84
Manganese	Mn	ppm	544	21		578	9.626	106.25			1757	58	1692.5	74.106	96.33	440	19	441.5	25.173	100.34
Nickel	Ni	ppm	42.9	3.7		44	0.816	102.56			75.4	1.5	76.5	1.000	101.46	39.5	2.3	41.25	1.893	104.43
Lead	Pb	ppm	150	17		141.25	6.021	94.17			132.8	1.1	120.5	3.317	90.74	183	8	169.25	8.261	92.49
Zinc	Zn	ppm	408	15		383.5	11.958	94.00			485.3	4.2	445	15.033	91.70	364	23	355.5	10.661	97.66
Silver	Au	ppb														1.22	0.14			
Gold	Ag	ppm									0.622	0.078	0.975	0.050	156.75					
Molybdenum	Mo	ppm									10.8	1.6	5.75	2.630	53.24	5.43	0.28	4.25	0.500	78.27
Aluminum	Al	%	6.1	0.18		6.155	0.288	100.90			8.41	0.22	7.6175	0.497	90.58	6.62	0.32	6.5975	0.316	99.66
Arsenic	As	ppm	17			21	2.309	123.53			45.3	1.8	57.25	3.948	126.38	26.2	1.5	36	6.683	137.40
Barium	Ba	ppm	413	13		422.5	95.350	102.30			397.4	3.2	460		115.75					
Beryllium	Be	ppm									3		3	0.000	100.00	1	0.2	1	0.000	100.00
Bismuth	Bi	ppm																		
Bromine	Br	ppm																		
Calcium	Ca	%	2.641	0.083		2.815	0.070	106.59			0.343	0.024	0.335	0.030	97.67	1.96	0.18	2.125	0.104	108.42
Cobalt	Co	ppm	13.57	0.43		13.5	1.291	99.48			27.76	0.58	27.25	3.594	98.16	11.5	0.3	13.5	0.577	117.39
Cesium	Cs	ppm	5.83	0.12		4.5	0.577	77.19			7.1									
Europium	Eu	ppm	1.31	0.038		1.2	0.183	91.60												
Hafnium	Hf	ppm	8.4	1.5		7.25	0.957	86.31			12.6		8.5	0.577	67.46					
Mercury	Hg	ppm									0.4474	0.0069				3.04	0.2			
Iridium	Ir_ppb	ppb																		
Potassium	K	%	2.001	0.041		2.0725	0.314	103.57			2.054	0.072	2.1725	0.126	105.77	1.24	0.05	1.23	0.243	99.19
Magnesium	Mg	%	1.2	0.018		1.1575	0.039	96.46			0.99	0.074	0.91	0.018	91.92	1.47	0.13	1.38	0.076	93.88
Sodium	Na	%	0.553	0.015		0.5975	0.034	108.05			0.681	0.02	0.7625	0.086	111.97	3.45	0.17	3.29	0.109	95.36
Antimony	Sb	ppm	3.07	0.32		3.4	0.141	110.75			5.6	0.24	5.825	0.126	104.02	11.3	2.6	12.7	0.535	112.39
Rubidium	Rb	ppm									127.7	8.8								
Scandium	Sc	ppm	11.26	0.19		11.5	0.258	102.13			25.9	1.1	23.825	0.222	91.99					
Selenium	Se	ppm									4.95	0.46				0.92	0.22			
Strontium	Sr	ppm									119.7	3	110	9.592	91.90	276	30	268.5	13.675	97.28
Tantalum	Ta	ppm																		
Titanium	Ti	%	0.457	0.02		0.425	0.024	93.00			0.884	0.082	0.74	0.110	83.71	0.443	0.032	0.445	0.024	100.45
Thorium	Th	ppm	9.07	0.16		9.175	0.737	101.16			20.51	0.96	20.2	1.447	98.49					
Uranium	U	ppm	3.09	0.13		4.03	0.115	130.53			10.4		7.425	1.857		3				
Vanadium	V	ppm	94.6	4							357.6	9.2	305.5	49.776	85.43	133	5	131	6.481	
Tungsten	W	ppm									6.2									
Yttrium	Y	ppm																		
Lanthanum	La	ppm									73.5	4.2	73.4	5.193	99.86					
Cerium	Ce	ppm	66.5	2		55.75	9.465	83.83			123.4	5.8	108.25	17.614	87.72					
Neodymium	Nd	ppm									56		45.25	10.372	80.80					
Samarium	Sm	ppm									10.8		9.825	0.650	90.97					
Tin	Sn	%									31.6	2.4				19.8	2.5			
Terbium	Tb	ppm																		
Ytterbium	Yb	ppm																		
Lutetium	Lu	ppm																		

**Table F-5. Quality Assurance / Quality Control Values from elemental analysis. Results are from 16 reference samples submitted as blind unknowns and were run with the samples from this study.**

# Appendix G

## Core Physical Properties

Sample ID	RESULTS							FOLK'S CLASS
	%H2O	Bulk Density	%GRAVEL	%SAND	%SILT	%CLAY	SHEPCLASS	
Core 2 0-29cm	73.50	1.20	0.00	14.41	37.38	48.21	Silty-Clay	
Core 2 80-92cm	22.14	1.97	0.00	60.67	28.96	10.37	Silty-Sand	
Core 2 116-124cm	15.88	2.14	19.33	68.17	8.38	4.13		gravelly muddy Sand
Core 3 0-12cm	65.11	1.28	0.00	34.86	27.28	37.87	Sand-Silt-Clay	
Core 3 12-20cm	26.00	1.88	41.49	53.11	2.79	2.61		sandy Gravel
Core 3 20-54cm	52.99	1.42	0.00	3.94	45.33	50.73	Silty-Clay	
Core 3 54-56cm	52.77	1.43	0.00	38.59	40.53	20.88	Sand-Silt-Clay	
Core 3 56-71cm	35.42	1.69	0.00	42.34	36.41	21.25	Sand-Silt-Clay	
Core 3 71-80cm	22.63	1.96	0.00	69.68	21.42	8.89	Silty-Sand	
Core 3 80-93cm	31.73	1.76	0.00	72.33	18.61	9.06	Silty-Sand	
Core 3 93-112cm	17.47	2.09	29.96	56.45	8.46	5.12		gravelly muddy Sand
Core 3 112-154.5cm	9.77	2.33	38.62	38.14	13.91	9.33		muddy sandy Gravel
Core 4 0-12cm	68.29	1.25	0.00	3.98	40.99	55.04	Silty-Clay	
Core 4 12-16cm	51.48	1.44	0.00	17.01	32.39	50.60	Silty-Clay	
Core 4 16-34cm	26.59	1.87	0.00	50.69	22.81	26.50	Sand-Silt-Clay	
Core 4 34-54cm	20.34	2.02	0.00	61.17	23.26	15.57	Silty-Sand	
Core 4 54-67cm	20.11	2.02	5.04	61.19	20.30	13.47		gravelly muddy Sand
Core 4 67-82.5cm	12.71	2.23	21.63	48.30	17.66	12.41		gravelly muddy Sand
Core 5 0-30cm	74.50	1.19	0.00	1.58	43.08	55.33	Silty-Clay	
Core 5 30-62cm	56.85	1.38	0.00	0.89	44.06	55.05	Silty-Clay	
Core 5 62-76cm	41.88	1.58	0.00	46.83	33.48	19.69	Silty-Sand	
Core 5 76-89cm	25.18	1.90	0.00	65.88	22.40	11.72	Silty-Sand	
Core 6 0-16cm	74.21	1.19	0.00	3.67	42.82	53.51	Silty-Clay	
Core 6 16-44cm	61.33	1.32	0.00	3.15	39.19	57.66	Silty-Clay	
Core 6 44-48cm	63.45	1.30	0.00	2.02	39.39	58.60	Silty-Clay	
Core 6 48-62cm	61.02	1.33	0.00	4.25	38.04	57.71	Silty-Clay	
Core 6 62-74cm	56.32	1.38	0.00	64.89	17.66	17.45	Silty-Sand	
Core 6 74-84cm	16.64	2.11	7.55	19.25	59.84	13.36		gravelly muddy Sand
Core 8 0-16cm	63.17	1.30	0.00	12.47	39.20	48.33	Silty-Clay	
Core 8 16-44cm	49.03	1.48	0.00	27.24	41.06	31.71	Sand-Silt-Clay	
Core 8 44-56cm	19.73	2.03	0.00	86.47	9.57	3.96	Sand	
Core 8 56-81cm	31.28	1.77	0.00	74.10	17.55	8.34	Silty-Sand	
Core 8 81-83cm	40.91	1.60	0.00	82.42	10.35	7.23	Sand	
Core 8 83-92cm	35.85	1.68	0.00	83.40	11.53	5.07	Sand	
Core 8 92-100cm	23.98	1.93	0.00	91.06	6.40	2.54	Sand	
Core 13 0-22cm	72.61	1.21	0.00	4.88	60.54	34.58	Clayey-Silt	
Core 13 22-40cm	52.11	1.43	0.00	17.50	48.72	33.78	Clayey-Silt	
Core 13 40-56cm	21.33	1.99	0.00	44.08	36.46	19.47	Silty-Sand	
Core 13 56-102cm	21.16	1.99	0.00	80.55	13.61	5.84	Sand	
Core 13 102-109.5cm	22.23	1.97	0.00	89.76	6.69	3.55	Sand	
Core 16 0-13cm	7.43	2.41	60.87	30.80	6.65	1.68		muddy sandy Gravel
Core 16 13-43cm	5.74	2.48	49.15	41.44	6.12	3.29		muddy sandy Gravel

Table G-1. Physical properties of the collected core samples. Shephard's classification is used for samples without a gravel component. Folk's classification is used for samples with gravel.

# Appendix H

## Analysis of Cores 9, 11, and 15

Further analysis of the sediments was desired after the initial results were reported. Three cores were identified for further analysis due to their spatial location, existing data, and depth. Cores 9, 11, and 15, were analyzed for textural properties by MGS and for 48 elements by Actlabs. These cores were collected along the three up-stream transects in the Lake (Figure A-1 ).

### **STUDY OBJECTIVES**

The objectives for this follow-on study were:

- 1) Further document physical and elemental characteristics of the sediment.
- 2) Identify any possible trends in the historical core sediments.

### **METHODOLOGY**

Methods and interpreted analyses of the results are the same as those used for the first set of samples.

### **RESULTS AND DISCUSSION**

Analytical results are presented in Table H-2 . Results of the Standard reference material (SRM) used for QA/QC are listed in Table H-3.

The cores penetrated sediments ranging from silts and clays at the top of the sediment column to coarse sand and gravel at depth (Table H-2 ). Because of the significant downcore changes in the textural character of the sediments, assessing any changes in chemistry in the sediment column was difficult even when using enrichment factors to normalize the elemental data.

Generally, the elemental analyses for the cores 9, 11, and 15 yielded concentrations similar to those reported for Core 4. However, the iron concentration reported for at least one sample from each core was less than 1% which was unexpected given that these iron-poor samples contained some silts and clays which usually are iron-rich. Enrichment factors using Fe as the normalizing element would be relatively high for other elements due to the low iron in these samples.



**Table H-1. Comparison of average EF values using Fe and Al, respectively for sediments analyzed in Cores 9, 11, and 15.**

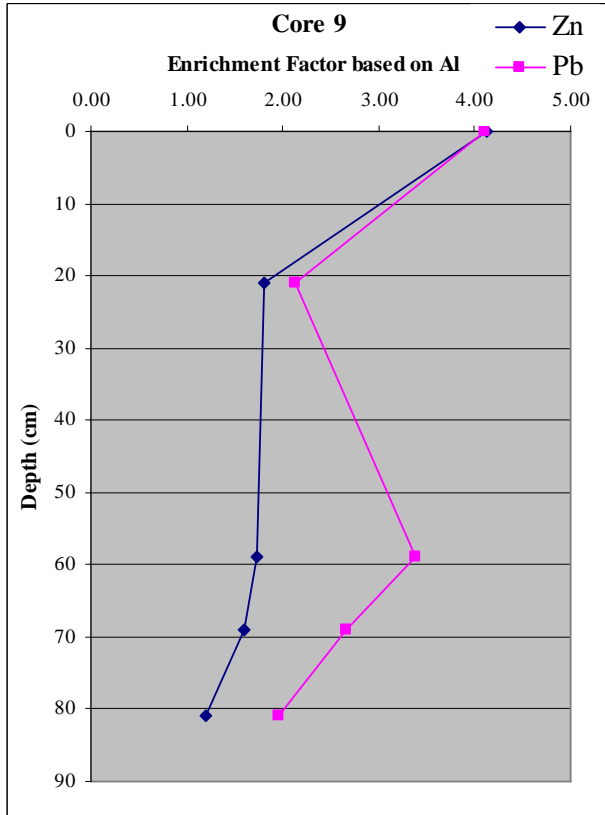
Element	Mean EF using		Element	Mean EF using	
	Al	Fe		Al	Fe
<b>P</b>	0.49	0.70	<b>Mg</b>	0.22	0.36
<b>Cr</b>	1.14	1.91	<b>Rb</b>	1.42	2.35
<b>Cu</b>	0.69	1.03	<b>Sb</b>	8.46	13.70
<b>Fe</b>	0.71		<b>Sc</b>	0.74	1.24
<b>Mn</b>	0.29	0.44	<b>Sr</b>	0.28	0.47
<b>Ni</b>	0.70	1.14	<b>Ti</b>	0.97	1.54
<b>Pb</b>	2.74	4.55	<b>Th</b>	2.30	3.96
<b>Zn</b>	2.18	3.48	<b>U</b>	3.27	5.77
<b>Al</b>		1.71	<b>V</b>	0.68	1.04
<b>As</b>	9.74	16.25	<b>Y</b>	1.43	2.64
<b>As- adjusted*</b>	7.79	13.00	<b>La</b>	2.07	3.53
<b>Ba</b>	1.46	2.43	<b>Ce</b>	1.96	3.34
<b>Co</b>	0.92	1.49	<b>Nd</b>	2.02	3.45
<b>Cs</b>	1.93	3.23	<b>Sm</b>	1.67	2.89
<b>Eu</b>	2.30	4.01	<b>Yb</b>	2.22	3.94
<b>Hf</b>	8.53	15.59	<b>Lu</b>	2.17	3.86

\* Arsenic concentrations were adjusted by multiplying reported concentration by 0.8 to correct for 125% recovery rate for reported by Actlabs.

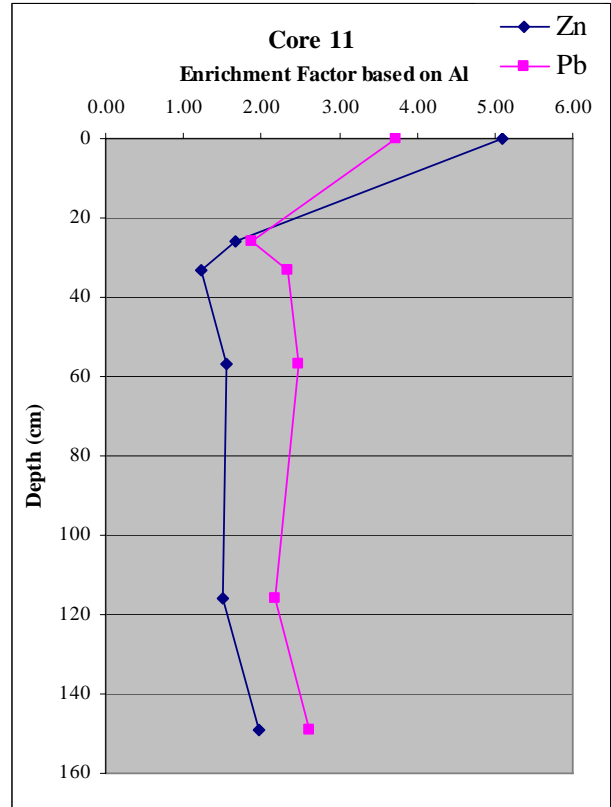
Enrichment factors calculated using Fe and Al were similar to those reported for first set of samples (Tables 2 and H-1). Elements having significantly high enrichment (i.e., >2 for both Fe and Al based EF values) include arsenic (As), lead (Pb), antimony (Sb), thorium (Th), uranium (U), zinc (Zn), lanthanum (La) and neodymium (Nd). Except for Pb and Zn, most elements do not show any downcore trend in enrichment. The high enrichment may reflect a natural regional abundance of the elements as they do not exhibit any significant downcore change. EFs tended to be higher in Core 15 which was collected in the upstream end of New Germany Lake and may have contained a higher percentage of unweathered parent rock.

Figures H-1 and H-2 show plots of the EF profiles for lead (Pb) and zinc (Zn) for Cores 9 and 11, respectively. Although the EFs were calculated using aluminum instead of iron, the downcore trends, particularly Zn, are similar to those for Core 4 (Figure 4). Plots suggest that sediments deeper than 21 cm [0.75 ft] in Core 9 and deeper than 33 cm [1 ft] in Core 11 are older than the early 1900s.

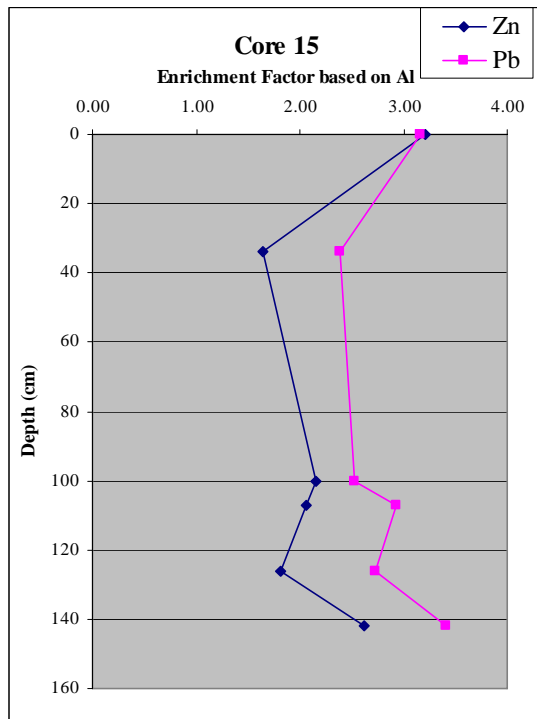
Plots of EF for Zn and Pb for Core 15 present a different picture (Figure H-3). EFs for both Zn and Pb do not show clear downcore trends. Core 15 was collected on the delta deposits at the upstream end of the lake. The delta area represents a higher energy deposition environment, subjected to storms and high flow events which would disturb and rework the sediments.



**Figure H-1 . Plot of EF (normalized using Al) for Zn and PB in core 9.**



**Figure H- 2 . Plots of EF (normalized using Al) for Zn and PB in core 11.**



**Figure H-3. Plot of EF (normalized using Al) for Zn and PB in core 15.**

**Table H-2 -. Results from textural and elemental analysis of samples from cores 9, 11 and 15.**

Sample ID and interval	%H2O	Bulk Density	%Gravel	%Sand	%Silt	%Clay	Shepard's Class	Folk's Class
Core 9 0-21cm	61.85	1.32	0.00	3.97	38.76	57.27	Silty-Clay	
Core 9 21-59cm	49.86	1.46	0.00	26.81	30.72	42.48	Sand-Silt-Clay	
Core 9 59-69cm	24.16	1.92	1.24	87.12	7.99	3.65		Slightly Gravelly Muddy Sand
Core 9 69-81cm	12.87	2.23	41.08	48.55	6.90	3.47		Muddy Sandy Gravel
Core 9 81-97cm	16.36	2.12	13.07	45.20	25.25	16.48		Gravelly Muddy Sand
Core 11 0-26cm	69.59	1.24	0.00	13.16	39.16	47.68	Silty-Clay	
Core 11 26-33cm	66.79	1.27	0.00	23.66	36.47	39.87	Sand-Silt-Clay	
Core 11 33-57cm	25.58	1.89	0.00	46.84	35.48	17.68	Silty-Sand	
Core 11 57-116cm	19.35	2.04	0.00	48.00	32.62	19.38	Silty-Sand	
Core 11 116-149cm	19.59	2.03	0.33	58.83	26.61	14.23		Slightly Gravelly Muddy Sand
Core 11 149-172cm	24.70	1.91	8.81	58.68	20.75	11.77		Gravelly Muddy Sand
Core 15 0-34cm	74.61	1.19	0.00	8.60	56.70	34.71	Clayey-Silt	
Core 15 34-100cm	29.88	1.80	0.00	29.65	48.23	22.11	Sand-Silt-Clay	
Core 15 100-107cm	28.91	1.82	0.00	79.05	13.08	7.86	Sand	
Core 15 107-126cm	15.13	2.16	50.19	34.51	9.90	5.40		Muddy Sandy Gravel
Core 15 126-142cm	10.11	2.32	49.22	35.87	10.02	4.88		Muddy Sandy Gravel
Core 15 142-145cm	9.85	2.33	32.44	47.71	8.25	11.60		Muddy Sandy Gravel

**Table H-2 –(cont.). Results from textural and elemental analysis of samples from cores 9, 11 and 15.**

	<b>Au</b>	<b>Ag</b>	<b>Cu</b>	<b>Cd</b>	<b>Mo</b>	<b>Pb</b>	<b>Ni</b>	<b>Zn</b>	<b>S</b>	<b>Al</b>	<b>As</b>	<b>Ba</b>	<b>Be</b>	<b>Bi</b>	<b>Br</b>	<b>Ca</b>
<b>Sample ID</b>	<b>ppb</b>	<b>ppm</b>	<b>ppm</b>	<b>ppm</b>	<b>ppm</b>	<b>ppm</b>	<b>ppm</b>	<b>ppm</b>	<b>%</b>	<b>%</b>	<b>ppm</b>	<b>ppm</b>	<b>ppm</b>	<b>ppm</b>	<b>ppm</b>	<b>%</b>
<b>Detection limit</b>	<b>2</b>	<b>0.3</b>	<b>1</b>	<b>0.3</b>	<b>1</b>	<b>3</b>	<b>1</b>	<b>1</b>	<b>0.01</b>	<b>0.01</b>	<b>0.5</b>	<b>50</b>	<b>1</b>	<b>2</b>	<b>0.5</b>	<b>0.01</b>
Core 9 0-21cm	< 2	< 0.3	26	2	2	44	51	248	0.14	7.06	13.2	650	4	< 2	8	0.16
Core 9 21-59cm	< 2	< 0.3	20	0.5	< 1	18	29	86	0.06	5.58	7.4	410	2	< 2	5.3	0.14
Core 9 59-69cm	< 2	< 0.3	3	< 0.3	< 1	7	6	20	0.01	1.36	2.6	130	< 1	< 2	1.2	0.05
Core 9 69-81cm	< 2	< 0.3	9	< 0.3	< 1	11	15	37	0.01	2.72	3.8	240	1	< 2	< 0.5	0.05
Core 9 81-97cm	< 2	0.3	10	0.6	< 1	12	21	41	0.03	4.02	4.5	220	2	< 2	1	0.16
Core 11 0-26cm	< 2	0.4	38	2.1	1	35	57	268	0.23	6.17	10.9	470	4	< 2	9.5	0.2
Core 11 26-33cm	< 2	0.3	18	0.5	< 1	13	26	65	0.13	4.57	6.6	340	2	< 2	4.1	0.43
Core 11 33-57cm	< 2	< 0.3	13	< 0.3	< 1	13	15	38	< 0.01	3.65	4.6	230	2	< 2	1.3	0.13
Core 11 57-116cm	< 2	< 0.3	7	< 0.3	< 1	12	16	42	< 0.01	3.18	4	260	1	< 2	0.9	0.12
Core 11 116-149cm	< 2	0.3	8	< 0.3	< 1	11	14	43	0.01	3.34	3.2	270	1	< 2	0.8	0.14
Core 11 149-172cm	< 2	< 0.3	16	0.4	< 1	15	23	63	0.07	3.77	6.5	270	2	< 2	< 0.5	0.18
Core 15 0-34cm	< 2	< 0.3	18	1.6	2	27	43	153	0.2	5.61	9.1	510	3	< 2	10.7	0.22
Core 15 34-100cm	< 2	0.4	15	0.5	< 1	16	27	62	0.05	4.42	7.7	270	2	< 2	2.1	0.18
Core 15 100-107cm	< 2	0.3	13	< 0.3	< 1	9	21	43	0.22	2.34	12	140	1	< 2	1.6	0.1
Core 15 107-126cm	< 2	0.4	23	0.3	< 1	20	34	79	0.13	4.49	23.4	200	2	< 2	< 0.5	0.08
Core 15 126-142cm	< 2	< 0.3	71	0.6	< 1	18	25	67	0.04	4.34	12.4	220	2	< 2	< 0.5	0.06
Core 15 142-145cm	< 2	0.4	19	0.9	< 1	14	27	60	0.02	2.7	9.8	320	2	< 2	< 0.5	0.06

**Table H-2 –(cont.). Results from textural and elemental analysis of samples from cores 9, 11 and 15.**

	Co	Cr	Cs	Eu	Fe	Hf	Hg	Ir	K	Mg	Mn	Na	P	Rb	Sb	Sc
<b>Sample ID</b>	<b>ppm</b>	<b>ppm</b>	<b>ppm</b>	<b>ppm</b>	<b>%</b>	<b>ppm</b>	<b>ppm</b>	<b>ppb</b>	<b>%</b>	<b>%</b>	<b>ppm</b>	<b>%</b>	<b>%</b>	<b>ppm</b>	<b>ppm</b>	<b>ppm</b>
<b>Detection limit</b>	<b>1</b>	<b>2</b>	<b>1</b>	<b>0.2</b>	<b>0.01</b>	<b>1</b>	<b>1</b>	<b>5</b>	<b>0.01</b>	<b>0.01</b>	<b>1</b>	<b>0.01</b>	<b>0.001</b>	<b>15</b>	<b>0.1</b>	<b>0.1</b>
<b>Core 9 0-21cm</b>	24	88	5	1.6	3.93	9	< 1	< 5	2.36	0.47	351	0.15	0.067	123	1.4	13.5
<b>Core 9 21-59cm</b>	11	57	4	1.2	2.17	10	< 1	< 5	1.64	0.35	220	0.12	0.036	72	0.7	9.1
<b>Core 9 59-69cm</b>	3	21	1	0.6	0.42	12	< 1	< 5	0.39	0.07	32	0.02	0.003	23	0.3	2.4
<b>Core 9 69-81cm</b>	4	35	2	1.3	1.46	12	< 1	< 5	1.13	0.12	34	0.04	0.02	40	0.7	4.9
<b>Core 9 81-97cm</b>	7	40	2	1.1	2.89	7	< 1	< 5	1.46	0.22	67	0.08	0.013	54	0.9	6.5
<b>Core 11 0-26cm</b>	22	76	5	1.5	3.65	9	< 1	< 5	2.05	0.42	347	0.13	0.063	107	0.9	11.5
<b>Core 11 26-33cm</b>	7	57	4	1.3	2.14	9	< 1	< 5	1.38	0.29	357	0.09	0.04	81	0.7	8.6
<b>Core 11 33-57cm</b>	5	48	3	1.3	1.14	14	< 1	< 5	0.95	0.19	92	0.08	0.013	48	0.5	6.8
<b>Core 11 57-116cm</b>	7	54	3	1.3	1.21	13	< 1	< 5	1.07	0.19	82	0.09	0.011	57	0.7	7.6
<b>Core 11 116-149cm</b>	6	48	2	1.2	0.8	12	< 1	< 5	0.94	0.18	66	0.07	0.008	46	0.5	6.5
<b>Core 11 149-172cm</b>	11	56	2	1.4	1.36	10	< 1	< 5	1.34	0.27	123	0.09	0.014	51	0.9	7.6
<b>Core 15 0-34cm</b>	18	66	5	1.3	2.48	9	< 1	< 5	1.69	0.32	234	0.11	0.079	94	0.8	10.4
<b>Core 15 34-100cm</b>	8	59	3	1.6	0.89	13	< 1	< 5	1.48	0.27	70	0.1	0.012	57	0.7	8.8
<b>Core 15 100-107cm</b>	12	29	1	1	0.73	9	< 1	< 5	0.7	0.13	28	0.04	0.009	35	0.6	4.5
<b>Core 15 107-126cm</b>	23	61	2	1.2	1.67	11	< 1	< 5	1.7	0.31	68	0.09	0.017	63	1.2	8.2
<b>Core 15 126-142cm</b>	12	60	2	1.2	3.74	10	< 1	< 5	1.58	0.27	284	0.06	0.035	57	1	7.7
<b>Core 15 142-145cm</b>	14	66	3	1.3	3.37	11	< 1	< 5	1.8	0.27	114	0.1	0.033	63	1	10

**Table H-2 –(cont.). Results from textural and elemental analysis of samples from cores 9, 11 and 15.**

	Se	Sr	Ta	Ti	Th	U	V	W	Y	La	Ce	Nd	Sm	Sn	Tb	Yb	Lu
<b>Sample ID</b>	<b>ppm</b>	<b>ppm</b>	<b>ppm</b>	<b>%</b>	<b>ppm</b>	<b>ppm</b>	<b>ppm</b>	<b>ppm</b>	<b>ppm</b>	<b>ppm</b>	<b>ppm</b>	<b>ppm</b>	<b>ppm</b>	<b>%</b>	<b>ppm</b>	<b>ppm</b>	<b>ppm</b>
<b>Detection limit</b>	<b>3</b>	<b>1</b>	<b>0.5</b>	<b>0.01</b>	<b>0.2</b>	<b>0.5</b>	<b>2</b>	<b>1</b>	<b>1</b>	<b>0.5</b>	<b>3</b>	<b>5</b>	<b>0.1</b>	<b>0.01</b>	<b>0.5</b>	<b>0.2</b>	<b>0.05</b>
Core 9 0-21cm	< 3	92	1.8	0.35	14.8	4.3	85	< 1	25	41.8	79	38	6	< 0.01	< 0.5	3.8	0.57
Core 9 21-59cm	< 3	66	< 0.5	0.14	10.7	4.3	39	< 1	23	30.5	55	23	4.5	< 0.01	< 0.5	3.2	0.45
Core 9 59-69cm	4	17	< 0.5	0.13	4.9	2.5	12	< 1	16	14.4	27	12	2.2	< 0.01	< 0.5	1.9	0.28
Core 9 69-81cm	< 3	61	< 0.5	0.22	7.6	3.4	37	< 1	17	30.2	60	34	5.1	< 0.01	< 0.5	2.3	0.35
Core 9 81-97cm	< 3	44	< 0.5	0.23	8.8	3	52	< 1	23	23.3	46	19	4	< 0.01	< 0.5	2.6	0.39
Core 11 0-26cm	< 3	84	< 0.5	0.37	13.1	5	72	< 1	26	36.4	66	38	5.6	< 0.01	< 0.5	3.2	0.49
Core 11 26-33cm	< 3	62	< 0.5	0.41	10	4.4	66	< 1	22	28.3	55	18	4.3	< 0.01	< 0.5	2.7	0.43
Core 11 33-57cm	4	45	0.7	0.11	10.5	5.1	14	< 1	23	29.5	54	29	4.7	< 0.01	< 0.5	3.3	0.49
Core 11 57-116cm	< 3	39	< 0.5	0.38	10.9	4	51	< 1	19	29.6	55	23	4.6	< 0.01	0.9	3.2	0.51
Core 11 116-149cm	< 3	39	< 0.5	0.14	9.3	3.8	18	< 1	22	25.4	48	22	4.3	< 0.01	0.8	3.1	0.54
Core 11 149-172cm	< 3	45	< 0.5	0.2	10	4	40	< 1	26	27.4	52	25	4.9	< 0.01	< 0.5	3	0.55
Core 15 0-34cm	< 3	68	< 0.5	0.39	11.3	4.5	66	< 1	24	32.5	63	26	5.1	< 0.01	< 0.5	3.3	0.55
Core 15 34-100cm	< 3	52	< 0.5	0.18	12.2	4.4	35	< 1	29	31.4	59	29	5.6	< 0.01	1.1	4.1	0.72
Core 15 100-107cm	< 3	25	< 0.5	0.18	6.9	2.9	21	< 1	19	19	37	23	3.4	< 0.01	< 0.5	2.2	0.38
Core 15 107-126cm	< 3	47	< 0.5	0.26	11.3	4.6	52	< 1	21	26.5	49	23	4.3	< 0.01	< 0.5	3.1	0.5
Core 15 126-142cm	< 3	46	0.9	0.16	10.1	4.4	38	< 1	21	25.6	52	23	4.2	< 0.01	< 0.5	2.7	0.54
Core 15 142-145cm	< 3	38	< 0.5	0.44	12	3.6	69	< 1	8	32.9	61	31	5.1	< 0.01	< 0.5	3.3	0.59

**Table H-3 . Quality Assurance / Quality Control Values from elemental analysis. Results are from 3 samples of NIST SRM 1646a, submitted as blind unknowns and were run with the second set of samples.**

NIST SRM 1646a- Estuarine Sediment								
Element	Symbol	Units	Certified/Referenced values		Actlab Results			
			Certified values	Std dev	Ave	Std dev	Detection limit	% Recovery
Gold	Au	ppb					2	
Silver	Ag	ppm			0.6		0.3	
Copper	Cu	ppm	10.01	0.34	10.33	0.58	1	103.2
Cadmium	Cd	ppm	0.148	0.007	0.450	0.071	0.3	304.1
Molybdenum	Mo	ppm	1.8		2	0	1	111.1
Lead	Pb	ppm	11.7	1.2	10	1	3	85.5
Nickel	Ni	ppm	23		25	2	1	108.7
Zinc	Zn	ppm	48.9	1.6	49	3	1	100.2
Sulfur	S	%	0.352	0.004	0.377	0.021	0.01	107.0
Aluminum	Al	%	2.297	0.018	2.043	0.652	0.01	89.0
Arsenic	As	ppm	6.23	0.21	8.03	0.40	0.5	128.9
Barium	Ba	ppm	210		200	17	50	95.2
Beryllium	Be	ppm			1	0	1	
Bismuth	Bi	ppm					2	
Bromine	Br	ppm			40.967	0.7371	0.5	
Calcium	Ca	%	0.519	0.02	0.547	0.110	0.01	105.3
Cobalt	Co	ppm	5		5.67	0.58	1	113.3
Chromium	Cr	ppm	40.9	1.9	50	2.6458	2	122.2
Cesium	Cs	ppm			1		1	
Europium	Eu	ppm			0.7333	0.2082	0.2	
Iron	Fe	%	2.008	0.039	1.94	0.04	0.01	96.6
Hafnium	Hf	ppm			13	0	1	
Potassium	K	%	0.864	0.016	1.107	0.081	0.01	128.1



**NIST SRM 1646a- Estuarine Sediment**

Element	Symbol	Units	Certified/Referenced values		Actlab Results			
			Certified values	Std dev	Ave	Std dev	Detection limit	% Recovery
<b>Magnesium</b>	<b>Mg</b>	%	<b>0.388</b>	<b>0.009</b>	0.417	0.049	<b>0.01</b>	<b>107.4</b>
<b>Manganese</b>	<b>Mn</b>	ppm	<b>234.5</b>	<b>2.8</b>	243.7	16.6	<b>1</b>	<b>103.9</b>
<b>Sodium</b>	<b>Na</b>	%	<b>0.741</b>	<b>0.017</b>	0.677	0.006	<b>0.01</b>	<b>91.3</b>
<b>Phosphorus</b>	<b>P</b>	%	<b>0.027</b>	<b>0.001</b>	0.030	0.001	<b>0.001</b>	<b>111.1</b>
<b>Rubidium</b>	<b>Rb</b>	ppm	<b>38</b>		35.5	2.1213	<b>15</b>	<b>93.4</b>
<b>Antimony</b>	<b>Sb</b>	ppm	<b>0.3</b>		0.3	0.1414	<b>0.1</b>	<b>100.0</b>
<b>Scandium</b>	<b>Sc</b>	ppm	<b>5</b>		4.8	0.1	<b>0.1</b>	<b>96.0</b>
<b>Selenium</b>	<b>Se</b>	ppm	<b>0.193</b>	<b>0.028</b>			<b>3</b>	
<b>Strontium</b>	<b>Sr</b>	ppm	<b>68</b>		69.3	13.3	<b>1</b>	<b>102.0</b>
<b>Titanium</b>	<b>Ti</b>	%	<b>0.456</b>	<b>0.021</b>	0.457	0.101	<b>0.01</b>	<b>100.1</b>
<b>Thorium</b>	<b>Th</b>	ppm	<b>5.8</b>		6.9333	0.1155	<b>0.2</b>	<b>119.5</b>
<b>Uranium</b>	<b>U</b>	ppm	<b>2</b>		2.5	0.6557	<b>0.5</b>	<b>125.0</b>
<b>Vanadium</b>	<b>V</b>	ppm	<b>44.84</b>	<b>0.76</b>	37.33	10.02	<b>2</b>	<b>83.3</b>
<b>Tungsten</b>	<b>W</b>	ppm					<b>1</b>	
<b>Yttrium</b>	<b>Y</b>	ppm			9.3333	2.8868	<b>1</b>	
<b>Lanthanum</b>	<b>La</b>	ppm	<b>17</b>		18.7	0.3464	<b>0.5</b>	<b>110.0</b>
<b>Cerium</b>	<b>Ce</b>	ppm	<b>34</b>		36	2.6458	<b>3</b>	<b>105.9</b>
<b>Neodymium</b>	<b>Nd</b>	ppm	<b>15</b>		18	2	<b>5</b>	<b>117.8</b>
<b>Samarium</b>	<b>Sm</b>	ppm			3	0	<b>0.1</b>	
<b>Tin</b>	<b>Sn</b>	%	<b>1</b>				<b>0.01</b>	
<b>Terbium</b>	<b>Tb</b>	ppm					<b>0.5</b>	
<b>Ytterbium</b>	<b>Yb</b>	ppm			1.6	0.1	<b>0.2</b>	
<b>Lutetium</b>	<b>Lu</b>	ppm			0.22	0.01	<b>0.05</b>	